

2-Fluorobenzoic acid, 3-ethylphenyl ester

Inchi:

InchiKey:

Formula:

SMILES:

Mol. weight [g/mol]:

InChI=1S/C15H13FO2/c1-2-11-6-5-7-12(10-11)18-15(17)13-8-3-4-9-14(13)16/h3-10H,2H

DIANMQIKLBLKOI-UHFFFAOYSA-N

C15H13FO2

CCc1cccc(OC(=O)c2ccccc2F)c1

244.26

Physical Properties

Property code	Value	Unit	Source
gf	-147.75	kJ/mol	Joback Method
hf	-343.72	kJ/mol	Joback Method
hfus	27.78	kJ/mol	Joback Method
hvap	63.20	kJ/mol	Joback Method
log10ws	-4.69		Crippen Method
logp	3.607		Crippen Method
mcvol	183.900	ml/mol	McGowan Method
pc	2438.65	kPa	Joback Method
rinpol	1932.00		NIST Webbook
tb	681.48	K	Joback Method
tc	910.11	K	Joback Method
tf	409.44	K	Joback Method
vc	0.702	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	472.85	J/mol×K	681.48	Joback Method
cpg	487.50	J/mol×K	719.58	Joback Method
cpg	501.06	J/mol×K	757.69	Joback Method
cpg	513.60	J/mol×K	795.79	Joback Method
cpg	525.13	J/mol×K	833.90	Joback Method
cpg	535.70	J/mol×K	872.00	Joback Method
cpg	545.35	J/mol×K	910.11	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U354705&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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