

# p-Pentylacetophenone

<b>Other names:</b>	p-n-Pentylacetophenone Acetophenone, 4'-pentyl- Ethanone, 1-(4-pentylphenyl)- 4'-Pentylacetophenone
<b>Inchi:</b>	InChI=1S/C13H18O/c1-3-4-5-6-12-7-9-13(10-8-12)11(2)14/h7-10H,3-6H2,1-2H3
<b>InchiKey:</b>	KBKGPMDADJLBEM-UHFFFAOYSA-N
<b>Formula:</b>	C13H18O
<b>SMILES:</b>	<chem>CCCCC1ccc(C(C)=O)cc1</chem>
<b>Mol. weight [g/mol]:</b>	190.28
<b>CAS:</b>	37593-02-5

## Physical Properties

Property code	Value	Unit	Source
gf	32.44	kJ/mol	Joback Method
hf	-199.17	kJ/mol	Joback Method
hfus	24.68	kJ/mol	Joback Method
hvap	54.22	kJ/mol	Joback Method
log10ws	-4.19		Crippen Method
logp	3.622		Crippen Method
mcvol	171.840	ml/mol	McGowan Method
pc	2295.91	kPa	Joback Method
tb	582.37	K	Joback Method
tc	788.27	K	Joback Method
tf	325.14	K	Joback Method
vc	0.661	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	416.78	J/molxK	582.37	Joback Method
cpg	432.70	J/molxK	616.69	Joback Method
cpg	447.73	J/molxK	651.00	Joback Method
cpg	461.91	J/molxK	685.32	Joback Method
cpg	475.25	J/molxK	719.63	Joback Method

cpg	487.80	J/molxK	753.95	Joback Method
cpg	499.59	J/molxK	788.27	Joback Method
dvisc	0.0022438	Paxs	325.14	Joback Method
dvisc	0.0011931	Paxs	368.01	Joback Method
dvisc	0.0007238	Paxs	410.88	Joback Method
dvisc	0.0004826	Paxs	453.75	Joback Method
dvisc	0.0003451	Paxs	496.63	Joback Method
dvisc	0.0002603	Paxs	539.50	Joback Method
dvisc	0.0002046	Paxs	582.37	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	405.50 ± 0.50	K	0.50	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C37593025&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C37593025&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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