

# 1-ethyl-cis-2-butylcyclopropane

<b>Inchi:</b>	InChI=1S/C9H18/c1-3-5-6-9-7-8(9)4-2/h8-9H,3-7H2,1-2H3/t8-,9+/m1/s1
<b>InchiKey:</b>	RNEGJSGKJQQNEI-BDAKNGLRSA-N
<b>Formula:</b>	C9H18
<b>SMILES:</b>	CCCCC1CC1CC
<b>Mol. weight [g/mol]:</b>	126.24

## Physical Properties

Property code	Value	Unit	Source
gf	77.94	kJ/mol	Joback Method
hf	-176.63	kJ/mol	Joback Method
hfus	18.27	kJ/mol	Joback Method
hvap	35.23	kJ/mol	Joback Method
log10ws	-3.00		Crippen Method
logp	3.223		Crippen Method
mcvol	126.810	ml/mol	McGowan Method
pc	2537.93	kPa	Joback Method
rinpol	888.00		NIST Webbook
rinpol	890.50		NIST Webbook
rinpol	890.50		NIST Webbook
rinpol	886.80		NIST Webbook
tb	407.39	K	Joback Method
tc	583.43	K	Joback Method
tf	204.89	K	Joback Method
vc	0.495	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	257.60	J/molxK	407.39	Joback Method
cpg	330.01	J/molxK	554.09	Joback Method
cpg	316.85	J/molxK	524.75	Joback Method
cpg	303.05	J/molxK	495.41	Joback Method
cpg	288.59	J/molxK	466.07	Joback Method
cpg	273.45	J/molxK	436.73	Joback Method

cpg	342.57	J/mol×K	583.43	Joback Method
dvisc	0.0003613	Paxs	407.39	Joback Method
dvisc	0.0003943	Paxs	373.64	Joback Method
dvisc	0.0004379	Paxs	339.89	Joback Method
dvisc	0.0004976	Paxs	306.14	Joback Method
dvisc	0.0005837	Paxs	272.39	Joback Method
dvisc	0.0007164	Paxs	238.64	Joback Method
dvisc	0.0009405	Paxs	204.89	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R136744&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R136744&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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