

# 1-Propene, 1-(1-methylethoxy)-, (E)-

Inchi:	InChI=1S/C6H12O/c1-4-5-7-6(2)3/h4-6H,1-3H3/b5-4+
InchiKey:	PLGKZYGGZOMZHL-SNAWJCMRSA-N
Formula:	C6H12O
SMILES:	CC=COC(C)C
Mol. weight [g/mol]:	100.16
CAS:	4188-65-2

## Physical Properties

Property code	Value	Unit	Source
gf	-27.58	kJ/mol	Joback Method
hf	-187.45	kJ/mol	Joback Method
hfus	9.16	kJ/mol	Joback Method
hvap	30.93	kJ/mol	Joback Method
log10ws	-1.88		Crippen Method
logp	1.945		Crippen Method
mcvol	96.970	ml/mol	McGowan Method
pc	3261.58	kPa	Joback Method
tb	362.82	K	Joback Method
tc	540.32	K	Joback Method
tf	159.53	K	Joback Method
vc	0.363	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	167.57	J/molxK	362.82	Joback Method
cpg	215.40	J/molxK	510.74	Joback Method
cpg	206.56	J/molxK	481.15	Joback Method
cpg	197.37	J/molxK	451.57	Joback Method
cpg	187.81	J/molxK	421.99	Joback Method
cpg	177.88	J/molxK	392.40	Joback Method
cpg	223.90	J/molxK	540.32	Joback Method
dvisc	0.0001756	Paxs	362.82	Joback Method
dvisc	0.0002340	Paxs	328.94	Joback Method

dvisc	0.0003332	Paxs	295.06	Joback Method
dvisc	0.0005198	Paxs	261.18	Joback Method
dvisc	0.0009258	Paxs	227.29	Joback Method
dvisc	0.0020189	Paxs	193.41	Joback Method
dvisc	0.0061309	Paxs	159.53	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4188652&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4188652&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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