

# 2-Adamantylamine, N,N-dimethyl-

<b>Other names:</b>	2-N,N-dimethylamino-adamantane
<b>Inchi:</b>	InChI=1S/C12H21N/c1-13(2)12-10-4-8-3-9(6-10)7-11(12)5-8/h8-12H,3-7H2,1-2H3
<b>InchiKey:</b>	LGQPYDNHZPYIRL-UHFFFAOYSA-N
<b>Formula:</b>	C12H21N
<b>SMILES:</b>	CN(C)C1C2CC3CC(C2)CC1C3
<b>Mol. weight [g/mol]:</b>	179.30

## Physical Properties

Property code	Value	Unit	Source
gf	315.67	kJ/mol	Joback Method
hf	-51.92	kJ/mol	Joback Method
hfus	24.30	kJ/mol	Joback Method
hvap	43.64	kJ/mol	Joback Method
log10ws	-2.24		Crippen Method
logp	2.373		Crippen Method
mcvol	157.340	ml/mol	McGowan Method
pc	2443.48	kPa	Joback Method
rinpol	1409.00		NIST Webbook
rinpol	1415.00		NIST Webbook
rinpol	1365.00		NIST Webbook
rinpol	1409.00		NIST Webbook
rinpol	1415.00		NIST Webbook
rinpol	1365.00		NIST Webbook
tb	501.55	K	Joback Method
tc	706.14	K	Joback Method
tf	299.29	K	Joback Method
vc	0.587	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	403.65	J/mol×K	501.55	Joback Method
cpg	426.62	J/mol×K	535.65	Joback Method
cpg	448.12	J/mol×K	569.75	Joback Method

cpg	468.24	J/mol×K	603.85	Joback Method
cpg	487.06	J/mol×K	637.94	Joback Method
cpg	504.69	J/mol×K	672.04	Joback Method
cpg	521.20	J/mol×K	706.14	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U374752&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U374752&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/49-106-0/2-Adamantylamine-N-N-dimethyl.pdf>

Generated by Cheméo on 2024-05-01 03:31:41.831822451 +0000 UTC m=+16823550.752399764.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.