

# 5-Octadecyne

<b>Inchi:</b>	InChI=1S/C18H34/c1-3-5-7-9-11-13-15-17-18-16-14-12-10-8-6-4-2/h3-9,11,13-18H2,1-2H
<b>InchiKey:</b>	LSZUIKJDOHTNDU-UHFFFAOYSA-N
<b>Formula:</b>	C18H34
<b>SMILES:</b>	CCCC#CCCCCCCCCCCCC
<b>Mol. weight [g/mol]:</b>	250.46
<b>CAS:</b>	71899-42-8

## Physical Properties

Property code	Value	Unit	Source
gf	303.48	kJ/mol	Joback Method
hf	-142.55	kJ/mol	Joback Method
hfus	45.50	kJ/mol	Joback Method
hvap	57.81	kJ/mol	Joback Method
log10ws	-7.15		Crippen Method
logp	6.491		Crippen Method
mcvol	255.880	ml/mol	McGowan Method
pc	1288.36	kPa	Joback Method
tb	620.24	K	Joback Method
tc	792.82	K	Joback Method
tf	265.40 ± 2.00	K	NIST Webbook
vc	1.006	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	685.49	J/molxK	620.24	Joback Method
cpg	705.29	J/molxK	649.00	Joback Method
cpg	724.24	J/molxK	677.77	Joback Method
cpg	742.38	J/molxK	706.53	Joback Method
cpg	759.73	J/molxK	735.29	Joback Method
cpg	776.31	J/molxK	764.06	Joback Method
cpg	792.16	J/molxK	792.82	Joback Method

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.56950e+01
Coeff. B	-5.33691e+03
Coeff. C	-1.04934e+02
Temperature range (K), min.	451.32
Temperature range (K), max.	618.91

## Sources

The Yaws Handbook of Vapor Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C71899428&Units=SI>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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