

# Cyclopentane, 1-bromo-2-fluoro-, cis-

<b>Other names:</b>	cis-1-Bromo-2-Fluorocyclopentane
<b>Inchi:</b>	InChI=1S/C5H8BrF/c6-4-2-1-3-5(4)7/h4-5H,1-3H2/t4-,5+/m0/s1
<b>InchiKey:</b>	GITCXTAKUHIPRH-CRCLSJGQSA-N
<b>Formula:</b>	C5H8BrF
<b>SMILES:</b>	FC1CCCC1Br
<b>Mol. weight [g/mol]:</b>	167.02
<b>CAS:</b>	51422-72-1

## Physical Properties

Property code	Value	Unit	Source
gf	-160.43	kJ/mol	Joback Method
hf	-276.17	kJ/mol	Joback Method
hfus	12.08	kJ/mol	Joback Method
hvap	32.29	kJ/mol	Joback Method
ie	10.10 ± 0.02	eV	NIST Webbook
log10ws	-2.32		Crippen Method
logp	2.272		Crippen Method
mcvol	89.720	ml/mol	McGowan Method
pc	4288.66	kPa	Joback Method
tb	389.84	K	Joback Method
tc	596.24	K	Joback Method
tf	213.16	K	Joback Method
vc	0.336	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	149.12	J/molxK	389.84	Joback Method
cpg	161.20	J/molxK	424.24	Joback Method
cpg	172.61	J/molxK	458.64	Joback Method
cpg	183.39	J/molxK	493.04	Joback Method
cpg	193.54	J/molxK	527.44	Joback Method
cpg	203.11	J/molxK	561.84	Joback Method
cpg	212.10	J/molxK	596.24	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C51422721&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C51422721&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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