

# Diethylmalonic acid, ethyl 3-phenoxybenzyl ester

Inchi:	InChI=1S/C22H26O5/c1-4-22(5-2,20(23)25-6-3)21(24)26-16-17-11-10-14-19(15-17)27-1
InchiKey:	MKJDLFQAMIMDLT-UHFFFAOYSA-N
Formula:	C22H26O5
SMILES:	CCOC(=O)C(CC)(CC)C(=O)OCc1cccc(Oc2ccccc2)c1
Mol. weight [g/mol]:	370.44

## Physical Properties

Property code	Value	Unit	Source
gf	-220.45	kJ/mol	Joback Method
hf	-666.39	kJ/mol	Joback Method
hfus	39.78	kJ/mol	Joback Method
hvap	89.21	kJ/mol	Joback Method
log10ws	-5.24		Crippen Method
logp	4.892		Crippen Method
mcvol	294.070	ml/mol	McGowan Method
pc	1491.89	kPa	Joback Method
rinpol	2546.00		NIST Webbook
tb	932.87	K	Joback Method
tc	1160.82	K	Joback Method
tf	572.03	K	Joback Method
vc	1.107	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	925.88	J/molxK	932.87	Joback Method
cpg	939.64	J/molxK	970.86	Joback Method
cpg	952.00	J/molxK	1008.85	Joback Method
cpg	963.01	J/molxK	1046.85	Joback Method
cpg	972.73	J/molxK	1084.84	Joback Method
cpg	981.21	J/molxK	1122.83	Joback Method
cpg	988.50	J/molxK	1160.82	Joback Method
dvisc	0.0002585	Paxs	572.03	Joback Method
dvisc	0.0001445	Paxs	632.17	Joback Method

dvisc	0.0000894	Paxs	692.31	Joback Method
dvisc	0.0000597	Paxs	752.45	Joback Method
dvisc	0.0000423	Paxs	812.59	Joback Method
dvisc	0.0000315	Paxs	872.73	Joback Method
dvisc	0.0000243	Paxs	932.87	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370229&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370229&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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