

# Propanal, 2,3-dihydroxy-, (S)-

<b>Other names:</b>	L-Glyceraldehyde L-(-)-glyceraldehyde
<b>Inchi:</b>	InChI=1S/C3H6O3/c4-1-3(6)2-5/h1,3,5-6H,2H2/t3-/m0/s1
<b>InchiKey:</b>	MNQZXJOMYWMBOU-VKHKMYHEASA-N
<b>Formula:</b>	C3H6O3
<b>SMILES:</b>	O=CC(O)CO
<b>Mol. weight [g/mol]:</b>	90.08
<b>CAS:</b>	497-09-6

## Physical Properties

Property code	Value	Unit	Source
chl	-1769.90	kJ/mol	NIST Webbook
chs	-1410.00	kJ/mol	NIST Webbook
gf	-401.22	kJ/mol	Joback Method
hf	-500.57	kJ/mol	Joback Method
hfus	10.47	kJ/mol	Joback Method
hvap	61.96	kJ/mol	Joback Method
log10ws	1.00		Crippen Method
logp	-1.462		Crippen Method
mcvol	66.440	ml/mol	McGowan Method
pc	6503.64	kPa	Joback Method
tb	500.62	K	Joback Method
tc	668.71	K	Joback Method
tf	272.21	K	Joback Method
vc	0.253	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	145.26	J/molxK	500.62	Joback Method
cpg	166.80	J/molxK	640.70	Joback Method
cpg	162.88	J/molxK	612.68	Joback Method
cpg	158.77	J/molxK	584.67	Joback Method
cpg	154.47	J/molxK	556.65	Joback Method

cpg	149.96	J/molxK	528.64	Joback Method
cpg	170.54	J/molxK	668.71	Joback Method
dvisc	0.0001203	Paxs	500.62	Joback Method
dvisc	0.0002424	Paxs	462.55	Joback Method
dvisc	0.0005539	Paxs	424.48	Joback Method
dvisc	0.0014895	Paxs	386.42	Joback Method
dvisc	0.0049726	Paxs	348.35	Joback Method
dvisc	0.0223150	Paxs	310.28	Joback Method
dvisc	0.1523958	Paxs	272.21	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C497096&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C497096&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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