

# N-Ethylidene t-butylamine

<b>Other names:</b>	t-Butylethyldeneamine Ethanamine, N-t-butyl- N-(Ethylidene)-2-methyl-2-propanamine (E) N-t-Butyl acetaldimine
<b>Inchi:</b>	InChI=1S/C6H13N/c1-5-7-6(2,3)4/h5H,1-4H3
<b>InchiKey:</b>	RPVBCVQKDPOII-UHFFFAOYSA-N
<b>Formula:</b>	C6H13N
<b>SMILES:</b>	CC=NC(C)(C)C
<b>Mol. weight [g/mol]:</b>	99.17
<b>CAS:</b>	7020-80-6

## Physical Properties

Property code	Value	Unit	Source
hf	-93.70	kJ/mol	Joback Method
hvap	30.97	kJ/mol	Joback Method
log10ws	-1.61		Crippen Method
logp	1.876		Crippen Method
mcvol	101.080	ml/mol	McGowan Method
pc	2829.33	kPa	Joback Method
tb	354.50 ± 0.50	K	NIST Webbook
tc	610.63	K	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7020806&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7020806&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

# Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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