

Methyl 2-hydroxy-4-methoxybenzoate, O-pentafluoropropionyl-

Inchi: InChI=1S/C12H9F5O5/c1-20-6-3-4-7(9(18)21-2)8(5-6)22-10(19)11(13,14)12(15,16)17/h3
InchiKey: ZATMWZCIBDRJDZ-UHFFFAOYSA-N
Formula: C12H9F5O5
SMILES: COC(=O)c1ccc(OC)cc1OC(=O)C(F)(F)C(F)(F)F
Mol. weight [g/mol]: 328.19

Physical Properties

Property code	Value	Unit	Source
gf	-1397.90	kJ/mol	Joback Method
hf	-1697.29	kJ/mol	Joback Method
hfus	27.43	kJ/mol	Joback Method
hvap	59.95	kJ/mol	Joback Method
log10ws	-3.54		Crippen Method
logp	2.585		Crippen Method
mcvol	185.780	ml/mol	McGowan Method
pc	2102.27	kPa	Joback Method
rinpol	1492.00		NIST Webbook
rinpol	1492.00		NIST Webbook
tb	675.49	K	Joback Method
tc	865.86	K	Joback Method
tf	450.80	K	Joback Method
vc	0.734	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	506.81	J/mol×K	675.49	Joback Method
cpg	517.78	J/mol×K	707.22	Joback Method
cpg	527.99	J/mol×K	738.95	Joback Method
cpg	537.46	J/mol×K	770.68	Joback Method
cpg	546.22	J/mol×K	802.41	Joback Method
cpg	554.26	J/mol×K	834.14	Joback Method
cpg	561.63	J/mol×K	865.86	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U374793&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rlnpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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