

# Benzoic acid, 2-hydroxy-3-nitro-

|                             |  |
|-----------------------------|--|
| <b>Other names:</b>         | Salicylic acid, 3-nitro-<br>2-Hydroxy-3-nitrobenzoic acid<br>3-Nitrosalicylic acid |
| <b>Inchi:</b>               | InChI=1S/C7H5NO5/c9-6-4(7(10)11)2-1-3-5(6)8(12)13/h1-3,9H,(H,10,11)                |
| <b>InchiKey:</b>            | WWWFHFGUOIQNJJC-UHFFFAOYSA-N   |
| <b>Formula:</b>             | C7H5NO5  |
| <b>SMILES:</b>              | O=C(O)c1cccc([N+](=O)[O-])c1O  |
| <b>Mol. weight [g/mol]:</b> | 183.12   |
| <b>CAS:</b>                 | 85-38-1  |

## Physical Properties

| Property code | Value   | Unit                 | Source         |
|---------------|---------|----------------------|----------------|
| gf            | -273.97 | kJ/mol               | Joback Method  |
| hf            | -415.63 | kJ/mol               | Joback Method  |
| hfus          | 30.37   | kJ/mol               | Joback Method  |
| hvap          | 87.14   | kJ/mol               | Joback Method  |
| log10ws       | -1.73   |                      | Crippen Method |
| logp          | 0.999   |                      | Crippen Method |
| mcvol         | 116.460 | ml/mol               | McGowan Method |
| pc            | 6318.86 | kPa                  | Joback Method  |
| tb            | 769.73  | K                    | Joback Method  |
| tc            | 1010.82 | K                    | Joback Method  |
| tf            | 573.67  | K                    | Joback Method  |
| vc            | 0.393   | m <sup>3</sup> /kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 302.18 | J/mol×K | 769.73          | Joback Method |
| cpg           | 308.59 | J/mol×K | 809.91          | Joback Method |
| cpg           | 314.62 | J/mol×K | 850.09          | Joback Method |
| cpg           | 320.35 | J/mol×K | 890.28          | Joback Method |
| cpg           | 325.89 | J/mol×K | 930.46          | Joback Method |
| cpg           | 331.30 | J/mol×K | 970.64          | Joback Method |

## Sources

|                        |   |
|------------------------|---|
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                   |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C85381&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C85381&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>                               |
| <b>Crippen Method:</b> | <a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>                       |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                   |

## Legend

|                 |   |
|-----------------|---|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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