

4-Hexylphenyl trifluoroacetate

Inchi:	InChI=1S/C14H17F3O2/c1-2-3-4-5-6-11-7-9-12(10-8-11)19-13(18)14(15,16)17/h7-10H,2
InchiKey:	YQQNYIXNCOSQFQ-UHFFFAOYSA-N
Formula:	C14H17F3O2
SMILES:	CCCCCCc1ccc(OC(=O)C(F)(F)F)cc1
Mol. weight [g/mol]:	274.28

Physical Properties

Property code	Value	Unit	Source
gf	-645.73	kJ/mol	Joback Method
hf	-949.11	kJ/mol	Joback Method
hfus	30.28	kJ/mol	Joback Method
hvap	55.11	kJ/mol	Joback Method
log10ws	-4.92		Crippen Method
logp	4.277		Crippen Method
mvol	197.110	ml/mol	McGowan Method
pc	1864.33	kPa	Joback Method
rinpol	1435.00		NIST Webbook
rinpol	1435.00		NIST Webbook
tb	622.25	K	Joback Method
tc	808.73	K	Joback Method
tf	362.83	K	Joback Method
vc	0.778	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	521.05	J/mol×K	622.25	Joback Method
cpg	535.79	J/mol×K	653.33	Joback Method
cpg	549.69	J/mol×K	684.41	Joback Method
cpg	562.77	J/mol×K	715.49	Joback Method
cpg	575.06	J/mol×K	746.57	Joback Method
cpg	586.61	J/mol×K	777.65	Joback Method
cpg	597.44	J/mol×K	808.73	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U373440&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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