

# Sebacic acid, 10-chlorodecyl octyl ester

**Inchi:** InChI=1S/C28H53ClO4/c1-2-3-4-5-15-20-25-32-27(30)22-17-12-8-9-13-18-23-28(31)33-2  
**InchiKey:** CDEDJKAMRYIXKT-UHFFFAOYSA-N  
**Formula:** C28H53ClO4  
**SMILES:** CCCCCCCCOC(=O)CCCCCCCCC(=O)OCCCCCCCCCCI  
**Mol. weight [g/mol]:** 489.17

## Physical Properties

Property code	Value	Unit	Source
gf	-294.89	kJ/mol	Joback Method
hf	-1126.59	kJ/mol	Joback Method
hfus	78.05	kJ/mol	Joback Method
hvap	100.62	kJ/mol	Joback Method
log10ws	-9.42		Crippen Method
logp	8.914		Crippen Method
mvol	432.500	ml/mol	McGowan Method
pc	674.30	kPa	Joback Method
rinpol	3528.00		NIST Webbook
rinpol	3528.00		NIST Webbook
tb	1030.05	K	Joback Method
tc	1285.34	K	Joback Method
tf	579.56	K	Joback Method
vc	1.700	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1503.95	J/molxK	1030.05	Joback Method
cpg	1525.75	J/molxK	1072.60	Joback Method
cpg	1545.36	J/molxK	1115.15	Joback Method
cpg	1562.86	J/molxK	1157.70	Joback Method
cpg	1578.35	J/molxK	1200.25	Joback Method
cpg	1591.94	J/molxK	1242.80	Joback Method
cpg	1603.70	J/molxK	1285.34	Joback Method
dvisc	0.0002392	Paxs	579.56	Joback Method

dvisc	0.0001116	Paxs	654.64	Joback Method
dvisc	0.0000609	Paxs	729.72	Joback Method
dvisc	0.0000372	Paxs	804.81	Joback Method
dvisc	0.0000247	Paxs	879.89	Joback Method
dvisc	0.0000175	Paxs	954.97	Joback Method
dvisc	0.0000130	Paxs	1030.05	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U355344&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U355344&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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