

# Benzeneacetic acid, 2-phenylethyl ester

<b>Other names:</b>	Acetic acid, phenyl-, phenethyl ester «beta»-Phenylethyl phenylacetate Benzylcarbiny «alpha»-toluate Phenylacetic acid, phenethyl ester Phenylethyl phenylacetate 2-Phenylethyl phenylacetate Phenethyl phenylacetate Phenethyl «alpha»-toluate 2-Phenylethyl «alpha»-toluate Phenylacetic acid, 2-phenylethyl ester NSC 6676
<b>Inchi:</b>	InChI=1S/C16H16O2/c17-16(13-15-9-5-2-6-10-15)18-12-11-14-7-3-1-4-8-14/h1-10H,11-
<b>InchiKey:</b>	ZOZIRNMDEZKZHM-UHFFFAOYSA-N
<b>Formula:</b>	C16H16O2
<b>SMILES:</b>	O=C(Cc1ccccc1)OCCc1ccccc1
<b>Mol. weight [g/mol]:</b>	240.30
<b>CAS:</b>	102-20-5

## Physical Properties

Property code	Value	Unit	Source
gf	74.74	kJ/mol	Joback Method
hf	-145.31	kJ/mol	Joback Method
hfus	28.07	kJ/mol	Joback Method
hvap	64.92	kJ/mol	Joback Method
log10ws	-3.59		Crippen Method
logp	3.015		Crippen Method
mcvol	196.220	ml/mol	McGowan Method
pc	2388.85	kPa	Joback Method
rinpol	1919.00		NIST Webbook
rinpol	1919.00		NIST Webbook
rinpol	1898.00		NIST Webbook
rinpol	1898.00		NIST Webbook
rinpol	1882.50		NIST Webbook
rinpol	1898.00		NIST Webbook
rinpol	1924.30		NIST Webbook
rinpol	1924.30		NIST Webbook
ripol	2618.00		NIST Webbook

ripol	2618.00		NIST Webbook
tb	695.13	K	Joback Method
tc	928.08	K	Joback Method
tf	395.08	K	Joback Method
vc	0.740	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	518.86	J/mol×K	695.13	Joback Method
cpg	587.75	J/mol×K	889.25	Joback Method
cpg	576.23	J/mol×K	850.43	Joback Method
cpg	563.64	J/mol×K	811.60	Joback Method
cpg	549.92	J/mol×K	772.78	Joback Method
cpg	535.01	J/mol×K	733.95	Joback Method
cpg	598.26	J/mol×K	928.08	Joback Method
dvisc	0.0001223	Paxs	695.13	Joback Method
dvisc	0.0001572	Paxs	645.12	Joback Method
dvisc	0.0002108	Paxs	595.11	Joback Method
dvisc	0.0002983	Paxs	545.11	Joback Method
dvisc	0.0004527	Paxs	495.10	Joback Method
dvisc	0.0007547	Paxs	445.09	Joback Method
dvisc	0.0014320	Paxs	395.08	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C102205&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C102205&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/49-648-9/Benzeneacetic-acid-2-phenylethyl-ester.pdf>

Generated by Cheméo on 2024-04-19 16:13:04.082566112 +0000 UTC m=+15832433.003143427.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.