

# «beta»-Chloropropionaldehyde diethyl acetal

<b>Other names:</b>	3-Chloro-1,1-diethoxypropane Propane, 3-chloro-1,1-diethoxy- 3-Chloropropionaldehyde diethyl acetal Propionaldehyde, 3-chloro-, diethyl acetal
<b>Inchi:</b>	InChI=1S/C7H15ClO2/c1-3-9-7(5-6-8)10-4-2/h7H,3-6H2,1-2H3
<b>InchiKey:</b>	NXHONHDWVLPPCS-UHFFFAOYSA-N
<b>Formula:</b>	C7H15ClO2
<b>SMILES:</b>	CCOC(CCCI)OCC
<b>Mol. weight [g/mol]:</b>	166.65
<b>CAS:</b>	35573-93-4

## Physical Properties

Property code	Value	Unit	Source
gf	-216.31	kJ/mol	Joback Method
hf	-473.27	kJ/mol	Joback Method
hfus	16.94	kJ/mol	Joback Method
hvap	39.99	kJ/mol	Joback Method
log10ws	-1.69		Crippen Method
logp	2.014		Crippen Method
mcvol	133.470	ml/mol	McGowan Method
pc	2621.78	kPa	Joback Method
tb	441.39	K	Joback Method
tc	616.85	K	Joback Method
tf	228.03	K	Joback Method
vc	0.506	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	270.37	J/molxK	441.39	Joback Method
cpg	324.29	J/molxK	587.61	Joback Method
cpg	314.21	J/molxK	558.36	Joback Method
cpg	303.77	J/molxK	529.12	Joback Method
cpg	292.98	J/molxK	499.88	Joback Method

cpg	281.85	J/molxK	470.63	Joback Method
cpg	334.02	J/molxK	616.85	Joback Method
dvisc	0.0002044	Paxs	441.39	Joback Method
dvisc	0.0002711	Paxs	405.83	Joback Method
dvisc	0.0003796	Paxs	370.27	Joback Method
dvisc	0.0005711	Paxs	334.71	Joback Method
dvisc	0.0009465	Paxs	299.15	Joback Method
dvisc	0.0017981	Paxs	263.59	Joback Method
dvisc	0.0041725	Paxs	228.03	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	357.20	K	3.30	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C35573934&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C35573934&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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