

Fumaric acid, ethyl 4-heptyl ester

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|-----------------------------|---|
| Inchi: | InChI=1S/C13H22O4/c1-4-7-11(8-5-2)17-13(15)10-9-12(14)16-6-3/h9-11H,4-8H2,1-3H3 |
| InchiKey: | DYMCTBBMZSGDAR-MDZDMXLPSA-N |
| Formula: | C13H22O4 |
| SMILES: | CCCC(CCC)OC(=O)C=CC(=O)OCC |
| Mol. weight [g/mol]: | 242.31 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -331.48 | kJ/mol | Joback Method |
| hf | -689.31 | kJ/mol | Joback Method |
| hfus | 31.68 | kJ/mol | Joback Method |
| hvap | 62.41 | kJ/mol | Joback Method |
| log10ws | -2.95 | | Crippen Method |
| logp | 2.618 | | Crippen Method |
| mvol | 204.610 | ml/mol | McGowan Method |
| pc | 1867.55 | kPa | Joback Method |
| rinpol | 1574.00 | | NIST Webbook |
| rinpol | 1574.00 | | NIST Webbook |
| tb | 653.14 | K | Joback Method |
| tc | 838.21 | K | Joback Method |
| tf | 360.51 | K | Joback Method |
| vc | 0.785 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 547.34 | J/molxK | 653.14 | Joback Method |
| cpg | 562.25 | J/molxK | 683.98 | Joback Method |
| cpg | 576.43 | J/molxK | 714.83 | Joback Method |
| cpg | 589.89 | J/molxK | 745.67 | Joback Method |
| cpg | 602.64 | J/molxK | 776.52 | Joback Method |
| cpg | 614.70 | J/molxK | 807.36 | Joback Method |
| cpg | 626.08 | J/molxK | 838.21 | Joback Method |
| dvisc | 0.0018471 | Paxs | 360.51 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0008679 | Paxs | 409.28 | Joback Method |
| dvisc | 0.0004789 | Paxs | 458.05 | Joback Method |
| dvisc | 0.0002963 | Paxs | 506.82 | Joback Method |
| dvisc | 0.0001995 | Paxs | 555.60 | Joback Method |
| dvisc | 0.0001431 | Paxs | 604.37 | Joback Method |
| dvisc | 0.0001079 | Paxs | 653.14 | Joback Method |

Sources

| | |
|------------------------|---|
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U348512&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |

Legend

| | |
|---------------------------------------|---|
| cp_g: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| g_f: | Standard Gibbs free energy of formation |
| h_f: | Enthalpy of formation at standard conditions |
| h_{fus}: | Enthalpy of fusion at standard conditions |
| h_{vap}: | Enthalpy of vaporization at standard conditions |
| log₁₀w_s: | Log ₁₀ of Water solubility in mol/l |
| log_p: | Octanol/Water partition coefficient |
| mc_{vol}: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rin_{pol}: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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