

3-Methyl-4-nitrobenzoic acid

Other names:	4-nitro-m-toluic acid benzoic acid, 3-methyl-4-nitro- m-Toluic acid, 4-nitro-
Inchi:	InChI=1S/C8H7NO4/c1-5-4-6(8(10)11)2-3-7(5)9(12)13/h2-4H,1H3,(H,10,11)
InchiKey:	XDTTUTIFWDAMIX-UHFFFAOYSA-N
Formula:	C8H7NO4
SMILES:	Cc1cc(C(=O)O)ccc1[N+](=O)[O-]
Mol. weight [g/mol]:	181.15
CAS:	3113-71-1

Physical Properties

Property code	Value	Unit	Source
gf	-120.56	kJ/mol	Joback Method
hf	-270.43	kJ/mol	Joback Method
hfus	26.79	kJ/mol	Joback Method
hsub	119.30 ± 2.50	kJ/mol	NIST Webbook
hvap	77.02	kJ/mol	Joback Method
log10ws	-2.66		Crippen Method
logp	1.601		Crippen Method
mcvol	124.680	ml/mol	McGowan Method
pc	4339.67	kPa	Joback Method
tb	716.97	K	Joback Method
tc	947.46	K	Joback Method
tf	485.74	K	Joback Method
vc	0.482	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	314.41	J/mol×K	716.97	Joback Method
cpg	322.66	J/mol×K	755.38	Joback Method
cpg	330.28	J/mol×K	793.80	Joback Method
cpg	337.28	J/mol×K	832.21	Joback Method
cpg	343.70	J/mol×K	870.63	Joback Method

cpg	349.56	J/mol×K	909.04	Joback Method
cpg	354.89	J/mol×K	947.46	Joback Method

Datasets

Mass density, kg/m3

Temperature, K - Crystal	Pressure, kPa - Crystal	Mass density, kg/m3 - Crystal
303.15	101.10	1528.6
Reference		https://www.doi.org/10.1016/j.jct.2016.07.023

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C3113711&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Solubility determination and thermodynamic modeling of	https://www.doi.org/10.1016/j.jct.2016.07.023
Solubility of 3-nitro-2-naphthol in binary solvent mixtures of	https://www.doi.org/10.1016/j.jct.2016.10.019
acid in binary solvent mixtures of	https://www.doi.org/10.1021/acs.jced.7b00431
2,4,6-trinitrophenol in binary solvent mixtures	
of 2,2,2-trifluoroethanol and 1,1,1-trichloroethane	
at 283.15, 298.15, and 318.15 K	
Monomers, Supramolecular Chemistry	
Experiments at temperatures between	
297.15 and 351.75 K	
thermodynamic modelling:	

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hsub:	Enthalpy of sublimation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient

mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rhos:	Solid Density
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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