

# Dimethylmalonic acid, neopentyl octyl ester

<b>Inchi:</b>	InChI=1S/C18H34O4/c1-7-8-9-10-11-12-13-21-15(19)18(5,6)16(20)22-14-17(2,3)4/h7-14
<b>InchiKey:</b>	XCGBELQNCMMJGF-UHFFFAOYSA-N
<b>Formula:</b>	C18H34O4
<b>SMILES:</b>	CCCCCCCCOC(=O)C(C)(C)C(=O)OCC(C)(C)C
<b>Mol. weight [g/mol]:</b>	314.46

## Physical Properties

Property code	Value	Unit	Source
gf	-361.48	kJ/mol	Joback Method
hf	-921.95	kJ/mol	Joback Method
hfus	33.12	kJ/mol	Joback Method
hvap	71.38	kJ/mol	Joback Method
log10ws	-4.60		Crippen Method
logp	4.506		Crippen Method
mcvol	279.360	ml/mol	McGowan Method
pc	1261.06	kPa	Joback Method
rinpol	1838.00		NIST Webbook
rinpol	1838.00		NIST Webbook
tb	757.36	K	Joback Method
tc	944.92	K	Joback Method
tf	441.78	K	Joback Method
vc	1.069	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	852.64	J/molxK	757.36	Joback Method
cpg	931.76	J/molxK	913.66	Joback Method
cpg	917.81	J/molxK	882.40	Joback Method
cpg	902.97	J/molxK	851.14	Joback Method
cpg	887.18	J/molxK	819.88	Joback Method
cpg	870.42	J/molxK	788.62	Joback Method
cpg	944.84	J/molxK	944.92	Joback Method
dvisc	0.0000473	Paxs	757.36	Joback Method

dvisc	0.0000649	Paxs	704.76	Joback Method
dvisc	0.0000939	Paxs	652.17	Joback Method
dvisc	0.0001449	Paxs	599.57	Joback Method
dvisc	0.0002429	Paxs	546.97	Joback Method
dvisc	0.0004547	Paxs	494.38	Joback Method
dvisc	0.0009881	Paxs	441.78	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U361749&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U361749&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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