Ammonium thiocyanate

Other names: ammonium isothiocyanate

> ammonium rhodanate ammonium rhodanid ammonium sulfocyanate

Inchi: InChI=1S/CHNS.H3N/c2-1-3;/h3H;1H3 InchiKey: SOIFLUNRINLCBN-UHFFFAOYSA-N

Formula: CH4N2S

SMILES: N#C[S-].[NH4+]

Mol. weight [g/mol]: 76.12 CAS: 1762-95-4

Physical Properties

Property code	Value	Unit	Source
hsub	133.90	kJ/mol	NIST Webbook
SS	140.20	J/mol×K	NIST Webbook

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cps	125.40	J/mol×K	298.15	NIST Webbook

Sources

Separation of azeotropic mixtures Separation of azeotropic mixtures using high ionicity ionic liquid based so paring reported acquisity in the distribution of some Thiocyanate Salts in Coaqueous Selutions of Notice of Notice and Speed of Sound Measurements:

https://www.doi.org/10.1016/j.jct.2006.04
https://www.doi.org/10.1016/j.jct.2012.06
https://www.doi.org/10.1016/j.jct.2012.06
https://www.doi.org/10.1021/je800562h
https://www.doi.org/10.1021/je800621p
https://www.doi.org/10.1021/je800621p

https://www.doi.org/10.1016/j.fluid.2015.01.011

https://www.doi.org/10.1016/j.jct.2006.04.006

https://www.doi.org/10.1016/j.jct.2012.06.007

http://webbook.nist.gov/cgi/cbook.cgi?ID=C1762954&Units=SI

Legend

cps: Solid phase heat capacity

hsub: Enthalpy of sublimation at standard conditionsss: Solid phase molar entropy at standard conditions

Latest version available from:

https://www.chemeo.com/cid/50-007-8/Ammonium-thiocyanate.pdf

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