

# (3-Chloro-4-hexyloxy-phenyl)-acetic acid, methyl ester

Inchi:	InChI=1S/C15H21ClO3/c1-3-4-5-6-9-19-14-8-7-12(10-13(14)16)11-15(17)18-2/h7-8,10H,
InchiKey:	KXLBCBBKRMVEX-UHFFFAOYSA-N
Formula:	C15H21ClO3
SMILES:	CCCCCOc1ccc(CC(=O)OC)cc1Cl
Mol. weight [g/mol]:	284.78

## Physical Properties

Property code	Value	Unit	Source
gf	-182.28	kJ/mol	Joback Method
hf	-532.10	kJ/mol	Joback Method
hfus	36.04	kJ/mol	Joback Method
hvap	68.53	kJ/mol	Joback Method
log10ws	-4.45		Crippen Method
logp	4.015		Crippen Method
mcvol	224.000	ml/mol	McGowan Method
pc	1787.88	kPa	Joback Method
rinpol	2036.40		NIST Webbook
tb	715.38	K	Joback Method
tc	916.91	K	Joback Method
tf	434.58	K	Joback Method
vc	0.859	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	602.38	J/molxK	715.38	Joback Method
cpg	617.50	J/molxK	748.97	Joback Method
cpg	631.73	J/molxK	782.56	Joback Method
cpg	645.08	J/molxK	816.14	Joback Method
cpg	657.56	J/molxK	849.73	Joback Method
cpg	669.17	J/molxK	883.32	Joback Method
cpg	679.92	J/molxK	916.91	Joback Method
dvisc	0.0007768	Paxs	434.58	Joback Method
dvisc	0.0004628	Paxs	481.38	Joback Method

dvisc	0.0003022	Paxs	528.18	Joback Method
dvisc	0.0002115	Paxs	574.98	Joback Method
dvisc	0.0001562	Paxs	621.78	Joback Method
dvisc	0.0001204	Paxs	668.58	Joback Method
dvisc	0.0000960	Paxs	715.38	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R158006&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R158006&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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