

# Benzaldehyde, 3,5-dichloro-2-hydroxy-

<b>Other names:</b>	Salicylaldehyde, 3,5-dichloro- 2-Hydroxy-3,5-dichlorobenzaldehyde 3,5-Dichloro-2-hydroxybenzaldehyde 3,5-Dichlorosalicylaldehyde Benzaldehyde, 2-hydroxy, 3,5-dichloro
<b>Inchi:</b>	InChI=1S/C7H4Cl2O2/c8-5-1-4(3-10)7(11)6(9)2-5/h1-3,11H
<b>InchiKey:</b>	FABVMBDCVAJXMB-UHFFFAOYSA-N
<b>Formula:</b>	C7H4Cl2O2
<b>SMILES:</b>	O=Cc1cc(Cl)cc(Cl)c1O
<b>Mol. weight [g/mol]:</b>	191.01
<b>CAS:</b>	90-60-8

## Physical Properties

Property code	Value	Unit	Source
gf	-176.79	kJ/mol	Joback Method
hf	-268.59	kJ/mol	Joback Method
hfus	23.62	kJ/mol	Joback Method
hvap	63.28	kJ/mol	Joback Method
log10ws	-2.63		Crippen Method
logp	2.511		Crippen Method
mcvol	117.650	ml/mol	McGowan Method
pc	4822.53	kPa	Joback Method
rinpol	1388.00		NIST Webbook
rinpol	1370.00		NIST Webbook
rinpol	1370.00		NIST Webbook
rinpol	1355.00		NIST Webbook
rinpol	1377.00		NIST Webbook
rinpol	1403.00		NIST Webbook
rinpol	1388.00		NIST Webbook
rinpol	1388.00		NIST Webbook
ripol	2314.00		NIST Webbook
ripol	2364.00		NIST Webbook
ripol	2272.00		NIST Webbook
ripol	2314.00		NIST Webbook
ripol	2364.00		NIST Webbook
ripol	2303.00		NIST Webbook
ripol	2314.00		NIST Webbook

tb	600.34	K	Joback Method
tc	844.92	K	Joback Method
tf	433.67	K	Joback Method
vc	0.401	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	234.27	J/mol×K	600.34	Joback Method
cpg	241.01	J/mol×K	641.10	Joback Method
cpg	247.21	J/mol×K	681.87	Joback Method
cpg	252.93	J/mol×K	722.63	Joback Method
cpg	258.26	J/mol×K	763.39	Joback Method
cpg	263.28	J/mol×K	804.16	Joback Method
cpg	268.07	J/mol×K	844.92	Joback Method
dvisc	0.0007061	Paxs	433.67	Joback Method
dvisc	0.0004081	Paxs	461.45	Joback Method
dvisc	0.0002511	Paxs	489.23	Joback Method
dvisc	0.0001627	Paxs	517.00	Joback Method
dvisc	0.0001102	Paxs	544.78	Joback Method
dvisc	0.0000776	Paxs	572.56	Joback Method
dvisc	0.0000564	Paxs	600.34	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C90608&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C90608&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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