

Eudesma-3,5-dien-1 «beta»-ol

Inchi:	InChI=1S/C14H22O/c1-10(2)11-7-8-14(3)12(9-11)5-4-6-13(14)15/h4-5,9-11,13,15H,6-8H
InchiKey:	OPUGJSIJRMAFLM-VNXPTHQBSA-N
Formula:	C14H22O
SMILES:	CC(C)C1C=C2C=CCC(O)C2(C)CC1
Mol. weight [g/mol]:	206.32

Physical Properties

Property code	Value	Unit	Source
gf	37.93	kJ/mol	Joback Method
hf	-269.85	kJ/mol	Joback Method
hfus	17.28	kJ/mol	Joback Method
hvap	63.35	kJ/mol	Joback Method
log10ws	-3.83		Crippen Method
logp	3.306		Crippen Method
mcvol	183.670	ml/mol	McGowan Method
pc	2419.50	kPa	Joback Method
rinsol	1706.00		NIST Webbook
ripol	2327.00		NIST Webbook
tb	640.89	K	Joback Method
tc	850.89	K	Joback Method
tf	348.86	K	Joback Method
vc	0.683	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	519.34	J/mol×K	640.89	Joback Method
cpg	537.53	J/mol×K	675.89	Joback Method
cpg	554.73	J/mol×K	710.89	Joback Method
cpg	571.07	J/mol×K	745.89	Joback Method
cpg	586.67	J/mol×K	780.89	Joback Method
cpg	601.68	J/mol×K	815.89	Joback Method
cpg	616.21	J/mol×K	850.89	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R229693&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cp_g:	Ideal gas heat capacity
g_f:	Standard Gibbs free energy of formation
h_f:	Enthalpy of formation at standard conditions
h_{fus}:	Enthalpy of fusion at standard conditions
h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀ws:	Log ₁₀ of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
pc:	Critical Pressure
ri_{npol}:	Non-polar retention indices
ri_{pol}:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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