

# Benzene, 1-ethenyl-4-methyl-

<b>Other names:</b>	1-Ethenyl-4-methylbenzene 1-METHYL-4-ETHENYLBENZENE 1-Methyl-4-vinylbenzene 1-p-Tolylethene 4-METHYLSTYRENE 4-Vinyltoluene P-VINYLTOLUENE Styrene, p-methyl- p-Methylstyrene para-Methylstyrene
<b>Inchi:</b>	InChI=1S/C9H10/c1-3-9-6-4-8(2)5-7-9/h3-7H,1H2,2H3
<b>InchiKey:</b>	JLBJTVDPNSHNSKJ-UHFFFAOYSA-N
<b>Formula:</b>	C9H10
<b>SMILES:</b>	<chem>C=Cc1ccc(C)cc1</chem>
<b>Mol. weight [g/mol]:</b>	118.18
<b>CAS:</b>	622-97-9

## Physical Properties

Property code	Value	Unit	Source
af	0.3250		KDB
affp	861.70	kJ/mol	NIST Webbook
basg	832.80	kJ/mol	NIST Webbook
gf	215.52	kJ/mol	Joback Method
hcg	5042.56	kJ/mol	KDB
hcn	4822.897	kJ/mol	KDB
hf	121.40	kJ/mol	Joback Method
hfus	11.44	kJ/mol	Joback Method
hvap	37.90	kJ/mol	Joback Method
ie	8.20	eV	NIST Webbook
ie	8.10 ± 0.10	eV	NIST Webbook
log10ws	-2.77		Crippen Method
logp	2.638		Crippen Method
mcvol	109.610	ml/mol	McGowan Method
pc	3370.00	kPa	KDB
rinpol	980.00		NIST Webbook
rinpol	985.00		NIST Webbook
rinpol	155.70		NIST Webbook

rinpol	982.00		NIST Webbook
rinpol	985.00		NIST Webbook
rinpol	977.50		NIST Webbook
rinpol	978.00		NIST Webbook
rinpol	966.00		NIST Webbook
rinpol	155.70		NIST Webbook
rinpol	982.00		NIST Webbook
rinpol	979.00		NIST Webbook
rinpol	980.50		NIST Webbook
rinpol	965.90		NIST Webbook
rinpol	1000.00		NIST Webbook
rinpol	985.00		NIST Webbook
rinpol	978.60		NIST Webbook
rinpol	980.30		NIST Webbook
rinpol	978.60		NIST Webbook
rinpol	980.30		NIST Webbook
rinpol	977.50		NIST Webbook
rinpol	977.60		NIST Webbook
rinpol	977.90		NIST Webbook
rinpol	982.00		NIST Webbook
rinpol	966.00		NIST Webbook
rinpol	158.68		NIST Webbook
ripol	1348.60		NIST Webbook
ripol	1396.80		NIST Webbook
ripol	1385.20		NIST Webbook
ripol	1349.00		NIST Webbook
ripol	1349.00		NIST Webbook
tb	444.65 ± 3.00	K	NIST Webbook
tb	445.65 ± 4.00	K	NIST Webbook
tb	446.00	K	NIST Webbook
tb	444.95 ± 2.00	K	NIST Webbook
tb	442.20	K	KDB
tc	659.70	K	KDB
tf	239.00 ± 0.15	K	NIST Webbook
tf	235.35 ± 0.50	K	NIST Webbook
tf	202.85	K	NIST Webbook
tf	239.30 ± 0.40	K	NIST Webbook
tf	239.00	K	KDB
vc	0.412	m <sup>3</sup> /kmol	KDB
zc	0.2534370		KDB

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	266.56	J/molxK	646.99	Joback Method
cpg	199.98	J/molxK	433.66	Joback Method
cpg	212.82	J/molxK	469.22	Joback Method
cpg	224.92	J/molxK	504.77	Joback Method
cpg	236.32	J/molxK	540.33	Joback Method
cpg	247.04	J/molxK	575.88	Joback Method
cpg	257.10	J/molxK	611.44	Joback Method
dvisc	0.0002171	Paxs	433.66	Joback Method
dvisc	0.0020152	Paxs	228.37	Joback Method
dvisc	0.0010913	Paxs	262.58	Joback Method
dvisc	0.0006808	Paxs	296.80	Joback Method
dvisc	0.0004682	Paxs	331.01	Joback Method
dvisc	0.0003454	Paxs	365.23	Joback Method
dvisc	0.0002684	Paxs	399.44	Joback Method
hvapt	47.60	kJ/mol	347.00	NIST Webbook
hvapt	35.98	kJ/mol	442.20	KDB
rfi	1.53950		298.15	KDB

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.cheric.org/files/research/kdb/mol/mol724.mol">https://www.cheric.org/files/research/kdb/mol/mol724.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C622979&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C622979&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=724">https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=724</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>af:</b>	Acentric Factor
<b>affp:</b>	Proton affinity

<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hcg:</b>	Heat of Combustion, Gross form
<b>hcn:</b>	Heat of Combustion, Net Form
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rfi:</b>	Refractive Index
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume
<b>zc:</b>	Critical Compressibility

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