

# Diethylmalonic acid, isobutyl 2,4,4-trimethylpentyl ester

Inchi:	InChI=1S/C19H36O4/c1-9-19(10-2,16(20)22-12-14(3)4)17(21)23-13-15(5)11-18(6,7)8/h1
InchiKey:	SPMNKTXYOIMISB-UHFFFAOYSA-N
Formula:	C19H36O4
SMILES:	CCC(CC)(C(=O)OCC(C)C)C(=O)OCC(C)CC(C)(C)C
Mol. weight [g/mol]:	328.49

## Physical Properties

Property code	Value	Unit	Source
gf	-357.94	kJ/mol	Joback Method
hf	-953.15	kJ/mol	Joback Method
hfus	28.67	kJ/mol	Joback Method
hvap	72.83	kJ/mol	Joback Method
log10ws	-4.53		Crippen Method
logp	4.607		Crippen Method
mcvol	293.450	ml/mol	McGowan Method
pc	1192.35	kPa	Joback Method
rinsol	1790.00		NIST Webbook
tb	779.36	K	Joback Method
tc	971.10	K	Joback Method
tf	423.05	K	Joback Method
vc	1.113	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	912.49	J/molxK	779.36	Joback Method
cpg	930.90	J/molxK	811.32	Joback Method
cpg	948.21	J/molxK	843.27	Joback Method
cpg	964.46	J/molxK	875.23	Joback Method
cpg	979.69	J/molxK	907.19	Joback Method
cpg	993.96	J/molxK	939.15	Joback Method
cpg	1007.30	J/molxK	971.10	Joback Method
dvisc	0.0013351	Paxs	423.05	Joback Method
dvisc	0.0004975	Paxs	482.43	Joback Method

dvisc	0.0002302	Paxs	541.82	Joback Method
dvisc	0.0001240	Paxs	601.20	Joback Method
dvisc	0.0000747	Paxs	660.59	Joback Method
dvisc	0.0000489	Paxs	719.98	Joback Method
dvisc	0.0000341	Paxs	779.36	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369475&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369475&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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