

# 2-Pentanone, 4-hydroxy-4-methyl-

<b>Other names:</b>	(CH <sub>3</sub> ) <sub>2</sub> C(OH)CH <sub>2</sub> C(O)CH <sub>3</sub> 2-Hydroxy-2-methyl-4-pentanone 2-Methyl-2-pentanol-4-one 4-Hydroxy-2-keto-4-methylpentane 4-Hydroxy-4-methyl-2-pentanone 4-Hydroxy-4-methyl-pentan-2-on 4-Hydroxy-4-methylpentan-2-one 4-Hydroxy-4-methylpentanone-2 4-Idrossi-4-metil-pentan-2-one 4-Methyl-2-pentanon-4-ol 4-Methyl-4-hydroxy-2-pentanone 4-hydroxy-4-methyl-pentanone 4-methyl-2-pentanone-4-ol Acetylndimethylcarbinol Diacetonalcool Diacetonalcool Diacetonalkohol Diacetone Diacetone alcohol Diacetone-alcool Diketone alcohol NSC 9005 Pyranton Pyranton A Pyraton Tyranton UN 1148
<b>Inchi:</b>	InChI=1S/C6H12O2/c1-5(7)4-6(2,3)8/h8H,4H2,1-3H3
<b>InchiKey:</b>	SWXVUIWOUIDPGS-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>6</sub> H <sub>12</sub> O <sub>2</sub>
<b>SMILES:</b>	CC(=O)CC(C)(C)O
<b>Mol. weight [g/mol]:</b>	116.16
<b>CAS:</b>	123-42-2

## Physical Properties

Property code	Value	Unit	Source
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affp	822.90	kJ/mol	NIST Webbook
basg	791.10	kJ/mol	NIST Webbook
chl	-3483.30 ± 7.90	kJ/mol	NIST Webbook
gf	-263.26	kJ/mol	Joback Method
hf	-540.70	kJ/mol	NIST Webbook
hfl	-592.80 ± 7.70	kJ/mol	NIST Webbook
hfus	9.57	kJ/mol	Joback Method
hvap	52.13	kJ/mol	NIST Webbook
log10ws	-0.99		Crippen Method
logp	0.736		Crippen Method
mcvol	102.840	ml/mol	McGowan Method
pc	3810.39	kPa	Joback Method
rinpol	798.00		NIST Webbook
rinpol	811.00		NIST Webbook
rinpol	814.00		NIST Webbook
rinpol	810.00		NIST Webbook
rinpol	836.00		NIST Webbook
rinpol	818.00		NIST Webbook
rinpol	821.00		NIST Webbook
rinpol	844.00		NIST Webbook
rinpol	809.00		NIST Webbook
rinpol	841.30		NIST Webbook
rinpol	816.00		NIST Webbook
rinpol	811.00		NIST Webbook
rinpol	812.00		NIST Webbook
rinpol	847.00		NIST Webbook
rinpol	844.00		NIST Webbook
rinpol	850.00		NIST Webbook
rinpol	838.00		NIST Webbook
rinpol	838.00		NIST Webbook
rinpol	831.00		NIST Webbook
rinpol	839.00		NIST Webbook
rinpol	830.00		NIST Webbook
rinpol	834.00		NIST Webbook
rinpol	834.00		NIST Webbook
rinpol	834.00		NIST Webbook
rinpol	809.00		NIST Webbook
rinpol	811.00		NIST Webbook
rinpol	842.00		NIST Webbook
rinpol	840.00		NIST Webbook
rinpol	814.00		NIST Webbook
rinpol	810.00		NIST Webbook
rinpol	815.00		NIST Webbook
rinpol	838.00		NIST Webbook

rinpol	848.00	NIST Webbook
rinpol	848.00	NIST Webbook
rinpol	815.00	NIST Webbook
rinpol	784.00	NIST Webbook
rinpol	839.00	NIST Webbook
rinpol	842.00	NIST Webbook
rinpol	818.00	NIST Webbook
rinpol	821.00	NIST Webbook
rinpol	851.00	NIST Webbook
rinpol	820.00	NIST Webbook
rinpol	817.00	NIST Webbook
rinpol	817.00	NIST Webbook
rinpol	820.00	NIST Webbook
rinpol	821.00	NIST Webbook
rinpol	811.00	NIST Webbook
rinpol	838.00	NIST Webbook
rinpol	839.00	NIST Webbook
rinpol	820.00	NIST Webbook
rinpol	811.00	NIST Webbook
rinpol	798.00	NIST Webbook
ripol	1340.00	NIST Webbook
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ripol	1375.00	NIST Webbook
ripol	1338.00	NIST Webbook
ripol	1339.00	NIST Webbook
ripol	1339.00	NIST Webbook
ripol	1340.00	NIST Webbook
ripol	1390.00	NIST Webbook
ripol	1359.00	NIST Webbook
ripol	1376.00	NIST Webbook
ripol	1380.00	NIST Webbook

ripol	1358.00		NIST Webbook
ripol	1396.00		NIST Webbook
ripol	1358.00		NIST Webbook
ripol	1343.00		NIST Webbook
ripol	1345.00		NIST Webbook
ripol	1385.00		NIST Webbook
ripol	1343.00		NIST Webbook
tb	437.20	K	NIST Webbook
tb	439.20	K	NIST Webbook
tb	442.35	K	NIST Webbook
tb	442.35	K	NIST Webbook
tc	661.11	K	Joback Method
tf	230.35	K	NIST Webbook
tf	230.35	K	NIST Webbook
vc	0.386	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	226.86	J/molxK	479.50	Joback Method
cpg	270.68	J/molxK	630.84	Joback Method
cpg	262.86	J/molxK	600.57	Joback Method
cpg	254.59	J/molxK	570.30	Joback Method
cpg	245.85	J/molxK	540.04	Joback Method
cpg	236.61	J/molxK	509.77	Joback Method
cpg	278.06	J/molxK	661.11	Joback Method
dvisc	0.0081361	Paxs	305.38	Joback Method
dvisc	0.0028958	Paxs	340.20	Joback Method
dvisc	0.0012487	Paxs	375.02	Joback Method
dvisc	0.0006212	Paxs	409.85	Joback Method
dvisc	0.0003447	Paxs	444.67	Joback Method
dvisc	0.0298236	Paxs	270.55	Joback Method
dvisc	0.0002084	Paxs	479.50	Joback Method
hvapt	51.00	kJ/mol	368.00	NIST Webbook
hvapt	47.50	kJ/mol	344.50	NIST Webbook

srf	0.03	N/m	323.15	Thermo Physical Properties of 4-Hydroxy 4-Methyl Pentanone with Nitrobenzene or Ethyl Benzene at Temperatures of (303.15, 313.15, and 323.15) K and a Pressure of 0.1 MPa
srf	0.03	N/m	323.15	Thermo Physical Properties of 4-Hydroxy 4-Methyl Pentanone with Nitrobenzene or Ethyl Benzene at Temperatures of (303.15, 313.15, and 323.15) K and a Pressure of 0.1 MPa
srf	0.03	N/m	313.15	Thermo Physical Properties of 4-Hydroxy 4-Methyl Pentanone with Nitrobenzene or Ethyl Benzene at Temperatures of (303.15, 313.15, and 323.15) K and a Pressure of 0.1 MPa
srf	0.03	N/m	303.15	Thermo Physical Properties of 4-Hydroxy 4-Methyl Pentanone with Nitrobenzene or Ethyl Benzene at Temperatures of (303.15, 313.15, and 323.15) K and a Pressure of 0.1 MPa

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.78470e+01
Coeff. B	-5.91478e+03

Coeff. C	5.97000e+00
Temperature range (K), min.	330.88
Temperature range (K), max.	465.87

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C123422&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C123422&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Solubility Measurement and Thermodynamic Modeling of Benzoic Acid in Various Organic Solvents:</b>	<a href="https://www.doi.org/10.1021/acs.jced.8b00025">https://www.doi.org/10.1021/acs.jced.8b00025</a>
<b>Thermophysical Properties of 1,1,1-Trichloro-4-Methyl Pentanone with Water:</b>	<a href="https://www.doi.org/10.1021/je8002699">https://www.doi.org/10.1021/je8002699</a>
<b>Joback Method of Ethyl Benzene at Temperatures of 303.15, 313.15, and 323.15 K and a Pressure of 0.1 MPa:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>srf:</b>	Surface Tension
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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