

# Oxirane, 3-ethyl-2,2-dimethyl-

<b>Other names:</b>	Pentane, 2,3-epoxy-2-methyl- 2-Methyl-2-pentene oxide 2,3-Epoxy-2-methylpentane 2-Methyl-2,3-pentylene oxide 2-Methyl-2,3-epoxypentane 3-Ethyl-2,2-dimethyl oxirane
<b>Inchi:</b>	InChI=1S/C6H12O/c1-4-5-6(2,3)7-5/h5H,4H2,1-3H3
<b>InchiKey:</b>	OZVWXRGXOYUQEJ-UHFFFAOYSA-N
<b>Formula:</b>	C6H12O
<b>SMILES:</b>	CCC1OC1(C)C
<b>Mol. weight [g/mol]:</b>	100.16
<b>CAS:</b>	1192-22-9

## Physical Properties

Property code	Value	Unit	Source
gf	-38.93	kJ/mol	Joback Method
hf	-231.47	kJ/mol	Joback Method
hfus	12.18	kJ/mol	Joback Method
hvap	31.91	kJ/mol	Joback Method
log10ws	-1.54		Crippen Method
logp	1.574		Crippen Method
mcvol	90.410	ml/mol	McGowan Method
pc	3624.61	kPa	Joback Method
rinpol	700.60		NIST Webbook
rinpol	700.30		NIST Webbook
rinpol	701.40		NIST Webbook
rinpol	701.40		NIST Webbook
rinpol	707.00		NIST Webbook
rinpol	700.30		NIST Webbook
tb	365.94	K	Joback Method
tc	553.87	K	Joback Method
tf	221.55	K	Joback Method
vc	0.346	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	170.26	J/mol×K	365.94	Joback Method
cpg	183.37	J/mol×K	397.26	Joback Method
cpg	195.52	J/mol×K	428.58	Joback Method
cpg	206.80	J/mol×K	459.90	Joback Method
cpg	217.29	J/mol×K	491.22	Joback Method
cpg	227.05	J/mol×K	522.54	Joback Method
cpg	236.18	J/mol×K	553.87	Joback Method
hvapt	40.60	kJ/mol	337.50	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1192229&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1192229&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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