

# Propanoic acid, 3-(acetyloxy)-2-(hydroxymethyl)-, ethyl ester,

**Other names:**  
**(+)-**

Ethyl 3-(acetyloxy)-2-(hydroxymethyl)propanoate, (+)-

Ethyl propanoic acid, 3-(acetyloxy)-2-(hydroxymethyl)-

**Inchi:** InChI=1S/C8H14O5/c1-3-12-8(11)7(4-9)5-13-6(2)10/h7,9H,3-5H2,1-2H3

**InchiKey:** WFTLWYCUWKENDL-UHFFFAOYSA-N

**Formula:** C8H14O5

**SMILES:** CCOC(=O)C(CO)COC(C)=O

**Mol. weight [g/mol]:** 190.19

**CAS:** 125476-48-4

## Physical Properties

Property code	Value	Unit	Source
gf	-590.62	kJ/mol	Joback Method
hf	-855.56	kJ/mol	Joback Method
hfus	22.62	kJ/mol	Joback Method
hvap	68.00	kJ/mol	Joback Method
log10ws	0.08		Crippen Method
logp	-0.279		Crippen Method
mcvol	144.330	ml/mol	McGowan Method
pc	3107.10	kPa	Joback Method
rinpol	1308.00		NIST Webbook
rinpol	1308.00		NIST Webbook
tb	626.76	K	Joback Method
tc	805.81	K	Joback Method
tf	370.06	K	Joback Method
vc	0.544	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	373.00	J/molxK	626.76	Joback Method
cpg	383.17	J/molxK	656.60	Joback Method
cpg	392.88	J/molxK	686.44	Joback Method
cpg	402.14	J/molxK	716.29	Joback Method
cpg	410.94	J/molxK	746.13	Joback Method

cpg	419.27	J/molxK	775.97	Joback Method
cpg	427.12	J/molxK	805.81	Joback Method
dvisc	0.0032891	Paxs	370.06	Joback Method
dvisc	0.0012145	Paxs	412.84	Joback Method
dvisc	0.0005407	Paxs	455.63	Joback Method
dvisc	0.0002766	Paxs	498.41	Joback Method
dvisc	0.0001573	Paxs	541.19	Joback Method
dvisc	0.0000972	Paxs	583.98	Joback Method
dvisc	0.0000641	Paxs	626.76	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C125476484&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C125476484&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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