

# 1-Hexanethiol

<b>Other names:</b>	1-Hexylthiol Hexan-1-thiol Hexanethiol Hexyl mercaptan N-HEXYL MERCAPTAN N-HEXYLTHIOL USAF EK-4628 hexane-1-thiol n-Hexanethiol
<b>Inchi:</b>	InChI=1S/C6H14S/c1-2-3-4-5-6-7/h7H,2-6H2,1H3
<b>InchiKey:</b>	PMBXCGGQNSVESQ-UHFFFAOYSA-N
<b>Formula:</b>	C6H14S
<b>SMILES:</b>	CCCCCCS
<b>Mol. weight [g/mol]:</b>	118.24
<b>CAS:</b>	111-31-9

## Physical Properties

Property code	Value	Unit	Source
chl	-4788.50 ± 0.75	kJ/mol	NIST Webbook
gf	29.03	kJ/mol	Joback Method
hf	-129.30 ± 0.96	kJ/mol	NIST Webbook
hfl	-175.70 ± 0.92	kJ/mol	NIST Webbook
hfus	15.34	kJ/mol	Joback Method
hvap	44.80 ± 0.20	kJ/mol	NIST Webbook
hvap	46.40	kJ/mol	NIST Webbook
log10ws	-2.41		Crippen Method
logp	2.497		Crippen Method
mcvol	111.750	ml/mol	McGowan Method
pc	3341.24	kPa	Joback Method
rinpola	921.00		NIST Webbook
rinpola	927.30		NIST Webbook
rinpola	912.50		NIST Webbook
rinpola	920.90		NIST Webbook
rinpola	920.40		NIST Webbook
rinpola	922.00		NIST Webbook
rinpola	925.00		NIST Webbook
rinpola	918.00		NIST Webbook

rinpol	930.00		NIST Webbook
rinpol	929.00		NIST Webbook
rinpol	917.00		NIST Webbook
rinpol	904.00		NIST Webbook
rinpol	918.00		NIST Webbook
rinpol	904.00		NIST Webbook
rinpol	930.00		NIST Webbook
rinpol	921.00		NIST Webbook
rinpol	927.00		NIST Webbook
rinpol	912.00		NIST Webbook
rinpol	916.00		NIST Webbook
rinpol	903.00		NIST Webbook
rinpol	901.00		NIST Webbook
rinpol	908.00		NIST Webbook
rinpol	909.00		NIST Webbook
rinpol	909.00		NIST Webbook
rinpol	910.00		NIST Webbook
rinpol	920.00		NIST Webbook
rinpol	916.40		NIST Webbook
ripol	1145.00		NIST Webbook
ripol	1145.00		NIST Webbook
ripol	1153.20		NIST Webbook
ripol	1153.20		NIST Webbook
ripol	1165.00		NIST Webbook
ripol	1140.00		NIST Webbook
ripol	1165.00		NIST Webbook
ripol	1148.00		NIST Webbook
sl	343.21	J/mol×K	NIST Webbook
tb	420.00 ± 2.00	K	NIST Webbook
tb	422.85	K	KDB
tb	424.20	K	NIST Webbook
tb	425.75 ± 0.20	K	NIST Webbook
tc	588.15	K	Joback Method
tf	192.20 ± 0.30	K	NIST Webbook
tf	192.12 ± 0.30	K	NIST Webbook
tf	192.66 ± 0.02	K	NIST Webbook
tt	192.62 ± 0.02	K	NIST Webbook
vc	0.425	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	204.81	J/mol×K	399.54	Joback Method
cpg	266.84	J/mol×K	588.15	Joback Method
cpg	257.60	J/mol×K	556.72	Joback Method
cpg	247.94	J/mol×K	525.28	Joback Method
cpg	237.85	J/mol×K	493.85	Joback Method
cpg	227.30	J/mol×K	462.41	Joback Method
cpg	216.29	J/mol×K	430.98	Joback Method
cpl	230.68	J/mol×K	298.15	NIST Webbook
cpl	230.71	J/mol×K	298.15	NIST Webbook
hfust	18.03	kJ/mol	192.60	NIST Webbook
hfust	18.03	kJ/mol	192.60	NIST Webbook
hfust	18.01	kJ/mol	192.62	NIST Webbook
hvapt	42.40	kJ/mol	410.00	NIST Webbook
hvapt	43.90	kJ/mol	387.00	NIST Webbook
sfust	93.51	J/mol×K	192.62	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.55981e+01
Coeff. B	-4.00405e+03
Coeff. C	-5.95250e+01
Temperature range (K), min.	321.05
Temperature range (K), max.	448.77

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	8.11148e+01
Coeff. B	-8.11731e+03
Coeff. C	-9.63481e+00
Coeff. D	4.93320e-06
Temperature range (K), min.	192.62
Temperature range (K), max.	623.00

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.therich.org/files/research/kdb/mol/mol1834.mol">https://www.therich.org/files/research/kdb/mol/mol1834.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C111319&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C111319&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.therich.org/research/kdb/hcprop/showprop.php?cmpid=1834">https://www.therich.org/research/kdb/hcprop/showprop.php?cmpid=1834</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemo.com/doc/models/crippen_log10ws">https://www.chemo.com/doc/models/crippen_log10ws</a>

# Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>sl:</b>	Liquid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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