

# 5-Butylindan

<b>Inchi:</b>	InChI=1S/C13H18/c1-2-3-5-11-8-9-12-6-4-7-13(12)10-11/h8-10H,2-7H2,1H3
<b>InchiKey:</b>	UJGOUXAEHNPLGZ-UHFFFAOYSA-N
<b>Formula:</b>	C13H18
<b>SMILES:</b>	CCCCc1ccc2c(c1)CCC2
<b>Mol. weight [g/mol]:</b>	174.28

## Physical Properties

Property code	Value	Unit	Source
gf	220.19	kJ/mol	Joback Method
hf	-4.92	kJ/mol	Joback Method
hfus	19.75	kJ/mol	Joback Method
hvap	48.35	kJ/mol	Joback Method
log10ws	-4.19		Crippen Method
logp	3.518		Crippen Method
mcvol	159.410	ml/mol	McGowan Method
pc	2495.01	kPa	Joback Method
rinpol	1430.24		NIST Webbook
rinpol	1415.33		NIST Webbook
rinpol	1443.20		NIST Webbook
rinpol	1438.14		NIST Webbook
rinpol	1410.65		NIST Webbook
rinpol	1403.33		NIST Webbook
tb	544.89	K	Joback Method
tc	758.81	K	Joback Method
tf	309.91	K	Joback Method
vc	0.614	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	379.16	J/molxK	544.89	Joback Method
cpg	396.60	J/molxK	580.54	Joback Method
cpg	412.96	J/molxK	616.20	Joback Method
cpg	428.30	J/molxK	651.85	Joback Method

cpg	442.69	J/molxK	687.50	Joback Method
cpg	456.20	J/molxK	723.16	Joback Method
cpg	468.90	J/molxK	758.81	Joback Method
dvisc	0.0019150	Paxs	309.91	Joback Method
dvisc	0.0012415	Paxs	349.07	Joback Method
dvisc	0.0008784	Paxs	388.24	Joback Method
dvisc	0.0006622	Paxs	427.40	Joback Method
dvisc	0.0005234	Paxs	466.56	Joback Method
dvisc	0.0004291	Paxs	505.73	Joback Method
dvisc	0.0003619	Paxs	544.89	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R250632&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R250632&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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