

# Pentyl perfluoroheptanoate

<b>Other names:</b>	2,2,3,3,4,4,5,5,6,6,7,7,7-Tridecafluoro-heptanoic acid pentyl ester
<b>Inchi:</b>	InChI=1S/C12H11F13O2/c1-2-3-4-5-27-6(26)7(13,14)8(15,16)9(17,18)10(19,20)11(21,22)
<b>InchiKey:</b>	QFCYZBVUAQZNQJ-UHFFFAOYSA-N
<b>Formula:</b>	C12H11F13O2
<b>SMILES:</b>	CCCCCOC(=O)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)F
<b>Mol. weight [g/mol]:</b>	434.19

## Physical Properties

Property code	Value	Unit	Source
gf	-2699.25	kJ/mol	Joback Method
hf	-3137.74	kJ/mol	Joback Method
hfus	25.18	kJ/mol	Joback Method
hvap	33.06	kJ/mol	Joback Method
log10ws	-5.94		Crippen Method
logp	5.459		Crippen Method
mcpvol	210.390	ml/mol	McGowan Method
pc	1280.08	kPa	Joback Method
rinpol	991.00		NIST Webbook
rinpol	991.40		NIST Webbook
rinpol	991.00		NIST Webbook
tb	521.38	K	Joback Method
tc	657.28	K	Joback Method
tf	319.35	K	Joback Method
vc	0.899	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	586.69	J/molxK	521.38	Joback Method
cpg	600.06	J/molxK	544.03	Joback Method
cpg	612.60	J/molxK	566.68	Joback Method
cpg	624.36	J/molxK	589.33	Joback Method
cpg	635.36	J/molxK	611.98	Joback Method
cpg	645.65	J/molxK	634.63	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R70204&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R70204&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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