

# Methacrylic anhydride

<b>Other names:</b>	2-Propenoic acid, 2-methyl-, anhydride Methacrylic acid anhydride Methacryloyl anhydride 2-Methyl-2-propenoic acid anhydride
<b>Inchi:</b>	InChI=1S/C8H10O3/c1-5(2)7(9)11-8(10)6(3)4/h1,3H2,2,4H3
<b>InchiKey:</b>	DCUFMVPCXCSVNP-UHFFFAOYSA-N
<b>Formula:</b>	C8H10O3
<b>SMILES:</b>	<chem>C=C(C)C(=O)OC(=O)C(=C)C</chem>
<b>Mol. weight [g/mol]:</b>	154.16
<b>CAS:</b>	760-93-0

## Physical Properties

Property code	Value	Unit	Source
gf	-187.78	kJ/mol	Joback Method
hf	-334.55	kJ/mol	Joback Method
hfus	15.68	kJ/mol	Joback Method
hvap	48.12	kJ/mol	Joback Method
log10ws	-1.52		Crippen Method
logp	1.208		Crippen Method
mcvol	123.990	ml/mol	McGowan Method
pc	3145.56	kPa	Joback Method
tb	505.72	K	Joback Method
tc	705.07	K	Joback Method
tf	270.57	K	Joback Method
vc	0.477	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	262.40	J/molxK	505.72	Joback Method
cpg	272.95	J/molxK	538.94	Joback Method
cpg	282.99	J/molxK	572.17	Joback Method
cpg	292.52	J/molxK	605.39	Joback Method
cpg	301.56	J/molxK	638.62	Joback Method

cpg	310.12	J/mol×K	671.84	Joback Method
cpg	318.20	J/mol×K	705.07	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	360.20	K	1.70	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C760930&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C760930&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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