

Butane, 2,3-dimethyl-2-nitro-

Inchi:	InChI=1S/C6H13NO2/c1-5(2)6(3,4)7(8)9/h5H,1-4H3
InchiKey:	PKOVWKGRNZGQHY-UHFFFAOYSA-N
Formula:	C6H13NO2
SMILES:	CC(C)C(C)(C)[N+](=O)[O-]
Mol. weight [g/mol]:	131.17
CAS:	34075-28-0

Physical Properties

Property code	Value	Unit	Source
gf	35.59	kJ/mol	Joback Method
hf	-191.96	kJ/mol	Joback Method
hfus	11.72	kJ/mol	Joback Method
hvap	43.86	kJ/mol	Joback Method
log10ws	-2.29		Crippen Method
logp	1.698		Crippen Method
mcvol	112.820	ml/mol	McGowan Method
pc	3246.73	kPa	Joback Method
tb	446.65 ± 2.00	K	NIST Webbook
tc	705.02	K	Joback Method
tf	302.40 ± 1.00	K	NIST Webbook
vc	0.436	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	253.60	J/mol×K	484.85	Joback Method
cpg	266.82	J/mol×K	521.55	Joback Method
cpg	279.19	J/mol×K	558.24	Joback Method
cpg	290.74	J/mol×K	594.94	Joback Method
cpg	301.53	J/mol×K	631.63	Joback Method
cpg	311.58	J/mol×K	668.33	Joback Method
cpg	320.96	J/mol×K	705.02	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C34075280&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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