

# Diethylmalonic acid, 2,2-dichloroethyl tetradecyl ester

Inchi:	InChI=1S/C23H42Cl2O4/c1-4-7-8-9-10-11-12-13-14-15-16-17-18-28-21(26)23(5-2,6-3)22
InchiKey:	HKRLEKGIERSQBC-UHFFFAOYSA-N
Formula:	C23H42Cl2O4
SMILES:	CCCCCCCCCCCCCOC(=O)C(CC)(CC)C(=O)OCC(Cl)Cl
Mol. weight [g/mol]:	453.48

## Physical Properties

Property code	Value	Unit	Source
gf	-348.52	kJ/mol	Joback Method
hf	-1053.16	kJ/mol	Joback Method
hfus	58.36	kJ/mol	Joback Method
hvap	92.19	kJ/mol	Joback Method
log10ws	-7.85		Crippen Method
logp	7.384		Crippen Method
mcvol	374.290	ml/mol	McGowan Method
pc	874.80	kPa	Joback Method
rinpol	2723.00		NIST Webbook
tb	949.41	K	Joback Method
tc	1162.70	K	Joback Method
tf	540.55	K	Joback Method
vc	1.452	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1215.06	J/molxK	949.41	Joback Method
cpg	1289.79	J/molxK	1127.16	Joback Method
cpg	1277.24	J/molxK	1091.61	Joback Method
cpg	1263.55	J/molxK	1056.06	Joback Method
cpg	1248.66	J/molxK	1020.51	Joback Method
cpg	1232.52	J/molxK	984.96	Joback Method
cpg	1301.27	J/molxK	1162.70	Joback Method
dvisc	0.0000168	Paxs	949.41	Joback Method
dvisc	0.0000230	Paxs	881.27	Joback Method

dvisc	0.0000331	Paxs	813.12	Joback Method
dvisc	0.0000509	Paxs	744.98	Joback Method
dvisc	0.0000854	Paxs	676.84	Joback Method
dvisc	0.0001610	Paxs	608.69	Joback Method
dvisc	0.0003563	Paxs	540.55	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370789&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370789&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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