

# 1-Pentadecanol acetate

<b>Other names:</b>	Acetic acid, pentadecyl ester Pentadecyl acetate n-Pentadecyl acetate
<b>Inchi:</b>	InChI=1S/C17H34O2/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-16-19-17(2)18/h3-16H2,1-2H3
<b>InchiKey:</b>	AAAIZLQILBFRTJ-UHFFFAOYSA-N
<b>Formula:</b>	C17H34O2
<b>SMILES:</b>	CCCCCCCCCCCCCCCCOC(C)=O
<b>Mol. weight [g/mol]:</b>	270.45
<b>CAS:</b>	629-58-3

## Physical Properties

Property code	Value	Unit	Source
gf	-141.66	kJ/mol	Joback Method
hf	-639.01	kJ/mol	Joback Method
hfus	42.57	kJ/mol	Joback Method
hvap	62.59	kJ/mol	Joback Method
log10ws	-5.80		Crippen Method
logp	5.641		Crippen Method
mcvol	257.830	ml/mol	McGowan Method
pc	1271.87	kPa	Joback Method
rinpol	1907.40		NIST Webbook
rinpol	1857.00		NIST Webbook
rinpol	1843.00		NIST Webbook
rinpol	1850.00		NIST Webbook
tb	664.65	K	Joback Method
tc	832.26	K	Joback Method
tf	353.51	K	Joback Method
vc	1.012	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	732.28	J/molxK	664.65	Joback Method
cpg	750.62	J/molxK	692.59	Joback Method

cpg	768.17	J/mol×K	720.52	Joback Method
cpg	784.95	J/mol×K	748.46	Joback Method
cpg	800.97	J/mol×K	776.39	Joback Method
cpg	816.25	J/mol×K	804.33	Joback Method
cpg	830.80	J/mol×K	832.26	Joback Method
dvisc	0.0022120	Paxs	353.51	Joback Method
dvisc	0.0009695	Paxs	405.37	Joback Method
dvisc	0.0005124	Paxs	457.22	Joback Method
dvisc	0.0003083	Paxs	509.08	Joback Method
dvisc	0.0002038	Paxs	560.94	Joback Method
dvisc	0.0001445	Paxs	612.79	Joback Method
dvisc	0.0001081	Paxs	664.65	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	2.12034e+01
Coeff. B	-7.76900e+03
Coeff. C	-1.18278e+02
Temperature range (K), min.	489.72
Temperature range (K), max.	607.14

## Sources

The Yaws Handbook of Vapor Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C629583&Units=SI>

## Legend

cpg: Ideal gas heat capacity

<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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