

# Glutaric acid, isobutyl 2,2,2-trichloroethyl ester

Inchi:	InChI=1S/C11H17Cl3O4/c1-8(2)6-17-9(15)4-3-5-10(16)18-7-11(12,13)14/h8H,3-7H2,1-2
InchiKey:	ITWOXHYBQPUWBS-UHFFFAOYSA-N
Formula:	C11H17Cl3O4
SMILES:	CC(C)COC(=O)CCCC(=O)OCC(Cl)(Cl)Cl
Mol. weight [g/mol]:	319.61

## Physical Properties

Property code	Value	Unit	Source
gf	-461.49	kJ/mol	Joback Method
hf	-821.22	kJ/mol	Joback Method
hfus	31.47	kJ/mol	Joback Method
hvap	69.86	kJ/mol	Joback Method
log10ws	-3.47		Crippen Method
logp	3.269		Crippen Method
mvol	217.450	ml/mol	McGowan Method
pc	1947.51	kPa	Joback Method
rinpol	1860.00		NIST Webbook
rinpol	1860.00		NIST Webbook
tb	712.28	K	Joback Method
tc	914.88	K	Joback Method
tf	435.23	K	Joback Method
vc	0.830	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	549.54	J/molxK	712.28	Joback Method
cpg	602.34	J/molxK	881.12	Joback Method
cpg	593.29	J/molxK	847.35	Joback Method
cpg	583.50	J/molxK	813.58	Joback Method
cpg	572.96	J/molxK	779.81	Joback Method
cpg	561.65	J/molxK	746.05	Joback Method
cpg	610.68	J/molxK	914.88	Joback Method
dvisc	0.0000960	Paxs	712.28	Joback Method

dvisc	0.0001270	Paxs	666.11	Joback Method
dvisc	0.0001751	Paxs	619.93	Joback Method
dvisc	0.0002543	Paxs	573.75	Joback Method
dvisc	0.0003943	Paxs	527.58	Joback Method
dvisc	0.0006651	Paxs	481.41	Joback Method
dvisc	0.0012533	Paxs	435.23	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U359345&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U359345&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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