

# 1,1'-Biphenyl, 2-chloro-

<b>Other names:</b>	1-Chloro-2-phenylbenzene 2-Chloro-1,1'-biphenyl 2-Chlorobiphenyl 2-Chlorodiphenyl 2-Monochlorobiphenyl Biphenyl, 2-chloro- NSC 67354 PCB 1 o-Chlorobiphenyl o-Chlorodiphenyl
<b>Inchi:</b>	InChI=1S/C12H9Cl/c13-12-9-5-4-8-11(12)10-6-2-1-3-7-10/h1-9H
<b>InchiKey:</b>	LAXBNTIAOJWAOP-UHFFFAOYSA-N
<b>Formula:</b>	C12H9Cl
<b>SMILES:</b>	Clc1ccccc1-c1ccccc1
<b>Mol. weight [g/mol]:</b>	188.65
<b>CAS:</b>	2051-60-7

## Physical Properties

Property code	Value	Unit	Source
chl	-6072.07	kJ/mol	NIST Webbook
gf	253.42	kJ/mol	Joback Method
hf	154.84	kJ/mol	Joback Method
hfus	18.73	kJ/mol	Joback Method
hvap	72.10 ± 2.00	kJ/mol	NIST Webbook
ie	8.20 ± 0.02	eV	NIST Webbook
log10ws	-4.54		Estimated Solubility Method
log10ws	-4.54		Aqueous Solubility Prediction Method
logp	4.007		Crippen Method
mvol	144.660	ml/mol	McGowan Method
pc	3228.31	kPa	Joback Method
rinpol	1494.00		NIST Webbook
rinpol	1485.00		NIST Webbook
rinpol	1461.00		NIST Webbook
rinpol	1538.00		NIST Webbook
rinpol	1466.00		NIST Webbook

rinpol	1538.00		NIST Webbook
rinpol	1465.00		NIST Webbook
rinpol	1508.00		NIST Webbook
rinpol	1482.00		NIST Webbook
rinpol	1459.00		NIST Webbook
ss	254.30	J/molxK	NIST Webbook
ss	303.76	J/molxK	NIST Webbook
tb	547.20	K	NIST Webbook
tc	825.02	K	Joback Method
tf	305.30 ± 0.20	K	NIST Webbook
tt	304.94 ± 0.02	K	NIST Webbook
tt	304.94 ± 0.02	K	NIST Webbook
vc	0.540	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	356.46	J/molxK	739.92	Joback Method
cpg	366.74	J/molxK	782.47	Joback Method
cpg	304.63	J/molxK	569.73	Joback Method
cpg	319.35	J/molxK	612.28	Joback Method
cpg	332.84	J/molxK	654.83	Joback Method
cpg	345.19	J/molxK	697.37	Joback Method
cpg	376.10	J/molxK	825.02	Joback Method
cps	211.92	J/molxK	298.15	NIST Webbook
cps	208.03	J/molxK	298.15	NIST Webbook
dvisc	0.0002543	Paxs	528.15	Joback Method
dvisc	0.0002031	Paxs	569.73	Joback Method
dvisc	0.0018784	Paxs	320.28	Joback Method
dvisc	0.0010478	Paxs	361.85	Joback Method
dvisc	0.0006592	Paxs	403.43	Joback Method
dvisc	0.0004522	Paxs	445.00	Joback Method
dvisc	0.0003309	Paxs	486.58	Joback Method
hfust	14.54	kJ/mol	304.90	NIST Webbook
hfust	14.54	kJ/mol	304.90	NIST Webbook
hfust	14.52	kJ/mol	304.94	NIST Webbook
hfust	14.52	kJ/mol	304.94	NIST Webbook
hvapt	55.80	kJ/mol	475.00	NIST Webbook
hvapt	74.50	kJ/mol	328.00	NIST Webbook
hvapt	64.40	kJ/mol	368.00	NIST Webbook
hvapt	57.80	kJ/mol	474.50	NIST Webbook

hvapt	61.10	kJ/mol	451.50	NIST Webbook
sfust	47.61	J/mol×K	304.94	NIST Webbook
sfust	47.61	J/mol×K	304.94	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2051607&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2051607&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>
<b>Estimated Solubility Method:</b>	<a href="http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt">http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>ss:</b>	Solid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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