

# Propanoic acid, 3-bromo-2-chloro, 4-methylbutyl ester

Inchi:	InChI=1S/C9H16BrClO2/c1-7(2)4-3-5-13-9(12)8(11)6-10/h7-8H,3-6H2,1-2H3
InchiKey:	YEKNCNQJTIFJKB-UHFFFAOYSA-N
Formula:	C9H16BrClO2
SMILES:	CC(C)CCCOC(=O)C(Cl)CBr
Mol. weight [g/mol]:	271.58

## Physical Properties

Property code	Value	Unit	Source
gf	-211.51	kJ/mol	Joback Method
hf	-473.86	kJ/mol	Joback Method
hfus	24.29	kJ/mol	Joback Method
hvap	54.83	kJ/mol	Joback Method
log10ws	-2.91		Crippen Method
logp	2.968		Crippen Method
mcvol	174.850	ml/mol	McGowan Method
pc	2530.27	kPa	Joback Method
rinpol	1400.00		NIST Webbook
rinpol	1400.00		NIST Webbook
tb	584.32	K	Joback Method
tc	783.95	K	Joback Method
tf	323.07	K	Joback Method
vc	0.662	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	393.53	J/molxK	584.32	Joback Method
cpg	406.27	J/molxK	617.59	Joback Method
cpg	418.35	J/molxK	650.86	Joback Method
cpg	429.79	J/molxK	684.13	Joback Method
cpg	440.60	J/molxK	717.40	Joback Method
cpg	450.81	J/molxK	750.68	Joback Method
cpg	460.42	J/molxK	783.95	Joback Method
dvisc	0.0034276	Paxs	323.07	Joback Method

dvisc	0.0016085	Paxs	366.61	Joback Method
dvisc	0.0008864	Paxs	410.15	Joback Method
dvisc	0.0005477	Paxs	453.70	Joback Method
dvisc	0.0003681	Paxs	497.24	Joback Method
dvisc	0.0002638	Paxs	540.78	Joback Method
dvisc	0.0001987	Paxs	584.32	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R30278&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R30278&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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