

# N-butraldehyde phenylhydrazone

<b>Other names:</b>	Butyraldehyde, phenylhydrazone
<b>Inchi:</b>	InChI=1S/C10H14N2/c1-2-3-9-11-12-10-7-5-4-6-8-10/h4-9,12H,2-3H2,1H3/b11-9+
<b>InchiKey:</b>	TVUBYAHAPNBKPY-PKQNBQFBNSA-N
<b>Formula:</b>	C10H14N2
<b>SMILES:</b>	CCCC=NNc1ccccc1
<b>Mol. weight [g/mol]:</b>	162.23
<b>CAS:</b>	940-54-5

## Physical Properties

Property code	Value	Unit	Source
hf	122.49	kJ/mol	Joback Method
hvap	49.88	kJ/mol	Joback Method
log10ws	-2.89		Crippen Method
logp	2.884		Crippen Method
mcpvol	143.660	ml/mol	McGowan Method
pc	2659.77	kPa	Joback Method
tb	581.73	K	Joback Method
tc	805.37	K	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C940545&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C940545&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

Latest version available from:

<https://www.cheméo.com/cid/50-850-2/N-butraldehyde-phenylhydrazone.pdf>

Generated by Cheméo on 2024-04-26 21:29:20.432141125 +0000 UTC m=+16456209.352718447.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.