

p-Chloro-N,N-dimethylaniline

Inchi:	InChI=1S/C8H10ClN/c1-10(2)8-5-3-7(9)4-6-8/h3-6H,1-2H3
InchiKey:	IONGEXNDPXANJD-UHFFFAOYSA-N
Formula:	C8H10ClN
SMILES:	CN(C)c1ccc(Cl)cc1
Mol. weight [g/mol]:	155.63

Physical Properties

Property code	Value	Unit	Source
af	0.4080		Cheméo Relay (1.0)
affp	922.90	kJ/mol	NIST Webbook
basg	896.40	kJ/mol	NIST Webbook
gf	218.11	kJ/mol	Joback Method
hf	41.25	kJ/mol	Cheméo Relay (1.0)
hfus	17.35	kJ/mol	Joback Method
hvap	58.37	kJ/mol	Cheméo Relay (1.0)
ie	7.38	eV	NIST Webbook
ie	7.20 ± 0.10	eV	NIST Webbook
log10ws	-2.86		Cheméo Relay (1.0)
logp	2.406		Crippen Method
mcvol	122.040	ml/mol	McGowan Method
pc	3423.86	kPa	Joback Method
tb	504.20	K	NIST Webbook
tc	710.54	K	Cheméo Relay (1.0)
tf	342.19	K	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	232.64	J/mol×K	463.97	Joback Method
cpg	245.45	J/mol×K	499.82	Joback Method
cpg	257.46	J/mol×K	535.68	Joback Method
cpg	268.70	J/mol×K	571.53	Joback Method
cpg	279.22	J/mol×K	607.39	Joback Method
cpg	289.03	J/mol×K	643.24	Joback Method

Sources

Cheméo Relay (1.0):

Cheméo Relay (1.0, beta):

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C698691&Units=SI>

Joback Method:

https://en.wikipedia.org/wiki/Joback_method

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws

Legend

af:	Acentric Factor
affp:	Proton affinity
basg:	Gas basicity
cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point

Latest version available from:

<https://www.chemeo.com/cid/50-943-9/p-Chloro-N-N-dimethylaniline.pdf>

Generated by Cheméo on 2026-07-21 15:20:09.062866007 +0000 UTC m=+158304.904751380.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.