

# Glutaric acid, 2-ethoxyethyl hexyl ester

<b>Inchi:</b>	InChI=1S/C15H28O5/c1-3-5-6-7-11-19-14(16)9-8-10-15(17)20-13-12-18-4-2/h3-13H2,1-2
<b>InchiKey:</b>	FJSBXGYEVVZFDX-UHFFFAOYSA-N
<b>Formula:</b>	C15H28O5
<b>SMILES:</b>	CCCCCOC(=O)CCCC(=O)OCCOCC
<b>Mol. weight [g/mol]:</b>	288.38

## Physical Properties

Property code	Value	Unit	Source
gf	-497.42	kJ/mol	Joback Method
hf	-974.75	kJ/mol	Joback Method
hfus	41.37	kJ/mol	Joback Method
hvap	69.71	kJ/mol	Joback Method
log10ws	-2.91		Crippen Method
logp	2.860		Crippen Method
mcvol	242.960	ml/mol	McGowan Method
pc	1498.83	kPa	Joback Method
rinpola	1987.00		NIST Webbook
rinpola	1987.00		NIST Webbook
tb	717.60	K	Joback Method
tc	894.31	K	Joback Method
tf	425.36	K	Joback Method
vc	0.942	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	706.75	J/molxK	717.60	Joback Method
cpg	778.49	J/molxK	864.86	Joback Method
cpg	765.73	J/molxK	835.41	Joback Method
cpg	752.18	J/molxK	805.96	Joback Method
cpg	737.83	J/molxK	776.50	Joback Method
cpg	722.68	J/molxK	747.05	Joback Method
cpg	790.45	J/molxK	894.31	Joback Method
dvisc	0.0000807	Paxs	717.60	Joback Method

dvisc	0.0001048	Paxs	668.89	Joback Method
dvisc	0.0001419	Paxs	620.19	Joback Method
dvisc	0.0002024	Paxs	571.48	Joback Method
dvisc	0.0003082	Paxs	522.77	Joback Method
dvisc	0.0005119	Paxs	474.07	Joback Method
dvisc	0.0009550	Paxs	425.36	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U359622&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U359622&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/50-964-6/Glutaric-acid-2-ethoxyethyl-hexyl-ester.pdf>

Generated by Cheméo on 2024-04-28 05:43:47.073317083 +0000 UTC m=+16572275.993894396.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.