

# 4,7,12-Trimethylbenz[a]anthracene

<b>Other names:</b>	Benz(a)anthracene, 4,7,12-trimethyl-
<b>Inchi:</b>	InChI=1S/C21H18/c1-13-7-6-10-20-16(13)11-12-19-14(2)17-8-4-5-9-18(17)15(3)21(19)2
<b>InchiKey:</b>	PYUNTMZRNSDKA-UHFFFAOYSA-N
<b>Formula:</b>	C21H18
<b>SMILES:</b>	<chem>Cc1cccc2c1ccc1c(C)c3cccc3c(C)c12</chem>
<b>Mol. weight [g/mol]:</b>	270.37
<b>CAS:</b>	35187-24-7

## Physical Properties

Property code	Value	Unit	Source
gf	510.15	kJ/mol	Joback Method
hf	275.62	kJ/mol	Joback Method
hfus	33.30	kJ/mol	Joback Method
hvap	72.85	kJ/mol	Joback Method
ie	7.06 ± 0.03	eV	NIST Webbook
log10ws	-8.31		Crippen Method
logp	6.071		Crippen Method
mcvol	224.610	ml/mol	McGowan Method
pc	1987.66	kPa	Joback Method
tb	788.40	K	Joback Method
tc	1036.95	K	Joback Method
tf	513.55	K	Joback Method
vc	0.870	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	623.87	J/molxK	788.40	Joback Method
cpg	639.68	J/molxK	829.83	Joback Method
cpg	654.57	J/molxK	871.25	Joback Method
cpg	668.68	J/molxK	912.68	Joback Method
cpg	682.20	J/molxK	954.10	Joback Method
cpg	695.27	J/molxK	995.53	Joback Method
cpg	708.06	J/molxK	1036.95	Joback Method

dvisc	0.0013788	Paxs	513.55	Joback Method
dvisc	0.0011437	Paxs	559.36	Joback Method
dvisc	0.0009759	Paxs	605.17	Joback Method
dvisc	0.0008516	Paxs	650.98	Joback Method
dvisc	0.0007565	Paxs	696.78	Joback Method
dvisc	0.0006819	Paxs	742.59	Joback Method
dvisc	0.0006221	Paxs	788.40	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C35187247&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C35187247&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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