

# D-(+)-Talose, pentakis(trifluoroacetate), methyloxime (anti)

**Inchi:** InChI=1S/C17H10F15NO11/c1-39-33-2-4(41-9(35)14(21,22)23)6(43-11(37)16(27,28)29)  
**InchiKey:** IOALWQFXAPVULK-UHFFFAOYSA-N  
**Formula:** C17H10F15NO11  
**SMILES:** CON=CC(OC(=O)C(F)(F)F)C(OC(=O)C(F)(F)F)C(OC(=O)C(F)(F)F)C(COC(=O)C(F)(F)F)  
**Mol. weight [g/mol]:** 689.24

## Physical Properties

Property code	Value	Unit	Source
hf	-4674.73	kJ/mol	Joback Method
hvap	84.65	kJ/mol	Joback Method
log10ws	-4.25		Crippen Method
logp	2.620		Crippen Method
mcvol	325.690	ml/mol	McGowan Method
pc	923.86	kPa	Joback Method
rinpol	1145.50		NIST Webbook
rinpol	1145.50		NIST Webbook
tb	1040.05	K	Joback Method
tc	1307.10	K	Joback Method

## Sources

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>  
**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=U380237&Units=SI>  
**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci990307I>  
**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)  
**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

**hf:** Enthalpy of formation at standard conditions  
**hvap:** Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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