

# 2-methyl-1-heptanol

Inchi:	InChI=1S/C8H18O/c1-3-4-5-6-8(2)7-9/h8-9H,3-7H2,1-2H3
InchiKey:	QZESEQBMSFFHRY-UHFFFAOYSA-N
Formula:	C8H18O
SMILES:	CCCCC(C)CO
Mol. weight [g/mol]:	130.23
CAS:	60435-70-3

## Physical Properties

Property code	Value	Unit	Source
gf	-122.78	kJ/mol	Joback Method
hf	-365.96	kJ/mol	Joback Method
hfus	17.04	kJ/mol	Joback Method
hvap	49.69	kJ/mol	Joback Method
log10ws	-2.19		Crippen Method
logp	2.195		Crippen Method
mvol	129.450	ml/mol	McGowan Method
pc	2811.36	kPa	Joback Method
tb	448.55 ± 0.50	K	NIST Webbook
tb	448.15 ± 2.00	K	NIST Webbook
tc	637.10	K	Joback Method
tf	225.74	K	Joback Method
vc	0.496	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	290.23	J/mol×K	474.18	Joback Method
cpg	302.08	J/mol×K	501.33	Joback Method
cpg	313.47	J/mol×K	528.49	Joback Method
cpg	324.43	J/mol×K	555.64	Joback Method
cpg	334.96	J/mol×K	582.79	Joback Method
cpg	345.08	J/mol×K	609.94	Joback Method
cpg	354.79	J/mol×K	637.10	Joback Method
cpl	327.80	J/mol×K	303.15	NIST Webbook

cpl	313.00	J/mol×K	298.50	NIST Webbook
dvisc	0.1178389	Paxs	225.74	Joback Method
dvisc	0.0169844	Paxs	267.15	Joback Method
dvisc	0.0041171	Paxs	308.55	Joback Method
dvisc	0.0013956	Paxs	349.96	Joback Method
dvisc	0.0005948	Paxs	391.37	Joback Method
dvisc	0.0002984	Paxs	432.77	Joback Method
dvisc	0.0001689	Paxs	474.18	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.72636e+01
Coeff. B	-5.42114e+03
Coeff. C	-1.94400e+01
Temperature range (K), min.	338.78
Temperature range (K), max.	473.01

## Sources

The Yaws Handbook of Vapor

Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C60435703&Units=SI>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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