

# Dimethylpropanedioic acid dimethyl ester

**Inchi:** InChI=1S/C7H12O4/c1-7(2,5(8)10-3)6(9)11-4/h1-4H3  
**InchiKey:** CAFVGIPKHPBMJS-UHFFFAOYSA-N  
**Formula:** C7H12O4  
**SMILES:** COC(=O)C(C)(C)C(=O)OC  
**Mol. weight [g/mol]:** 160.17  
**CAS:** 6065-54-9

## Physical Properties

Property code	Value	Unit	Source
chl	-3611.10 ± 0.42	kJ/mol	NIST Webbook
gf	-456.94	kJ/mol	Joback Method
hf	-802.74 ± 0.79	kJ/mol	NIST Webbook
hfl	-858.43 ± 0.42	kJ/mol	NIST Webbook
hfus	12.05	kJ/mol	Joback Method
hvap	55.69 ± 0.79	kJ/mol	NIST Webbook
hvap	55.70	kJ/mol	NIST Webbook
log10ws	-0.24		Crippen Method
logp	0.359		Crippen Method
mcvol	124.370	ml/mol	McGowan Method
pc	3159.72	kPa	Joback Method
tb	508.91	K	Joback Method
tc	706.01	K	Joback Method
tf	315.39	K	Joback Method
vc	0.465	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	281.74	J/molxK	508.91	Joback Method
cpg	341.51	J/molxK	706.01	Joback Method
cpg	332.83	J/molxK	673.16	Joback Method
cpg	323.65	J/molxK	640.31	Joback Method
cpg	313.95	J/molxK	607.46	Joback Method
cpg	303.74	J/molxK	574.61	Joback Method

cpg	293.00	J/mol×K	541.76	Joback Method
cpl	259.30	J/mol×K	298.15	NIST Webbook
dvisc	0.0014058	Paxs	347.64	Joback Method
dvisc	0.0002384	Paxs	508.91	Joback Method
dvisc	0.0003088	Paxs	476.66	Joback Method
dvisc	0.0004154	Paxs	444.40	Joback Method
dvisc	0.0005852	Paxs	412.15	Joback Method
dvisc	0.0008739	Paxs	379.90	Joback Method
dvisc	0.0024923	Paxs	315.39	Joback Method
hvapt	55.60 ± 0.80	kJ/mol	292.50	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6065549&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6065549&amp;Units=SI</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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