

# Ethane, 1,1,2,2-tetrachloro-1-fluoro-

<b>Other names:</b>	1,1,2,2-Tetrachloro-1-fluoroethane 1,1,2,2-Tetrachlorofluoroethane 1-Fluoro-1,1,2,2-tetrachloroethane Freon 121
<b>Inchi:</b>	InChI=1S/C2HCl4F/c3-1(4)2(5,6)7/h1H
<b>InchiKey:</b>	LUBCGHUOCJOIJA-UHFFFAOYSA-N
<b>Formula:</b>	C2HCl4F
<b>SMILES:</b>	FC(Cl)(Cl)C(Cl)Cl
<b>Mol. weight [g/mol]:</b>	185.84
<b>CAS:</b>	354-14-3

## Physical Properties

Property code	Value	Unit	Source
gf	-276.17	kJ/mol	Joback Method
hf	-357.71	kJ/mol	Joback Method
hfus	9.87	kJ/mol	Joback Method
hvap	35.09	kJ/mol	Joback Method
log10ws	-2.83		Crippen Method
logp	2.891		Crippen Method
mcvol	89.770	ml/mol	McGowan Method
pc	4026.13	kPa	Joback Method
tb	389.70	K	NIST Webbook
tc	599.15	K	Joback Method
tf	219.99	K	Joback Method
vc	0.344	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	127.79	J/molxK	390.48	Joback Method
cpg	132.49	J/molxK	425.26	Joback Method
cpg	136.72	J/molxK	460.04	Joback Method
cpg	140.49	J/molxK	494.81	Joback Method
cpg	143.85	J/molxK	529.59	Joback Method

cpg	146.83	J/mol×K	564.37	Joback Method
cpg	149.46	J/mol×K	599.15	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.34598e+01
Coeff. B	-2.57576e+03
Coeff. C	-9.83740e+01
Temperature range (K), min.	293.92
Temperature range (K), max.	414.55

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C354143&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C354143&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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