

3,5-Difluorophenylacetic acid

Inchi:	InChI=1S/C8H6F2O2/c9-6-1-5(3-8(11)12)2-7(10)4-6/h1-2,4H,3H2,(H,11,12)
InchiKey:	IGGNSAVLXJKCNH-UHFFFAOYSA-N
Formula:	C8H6F2O2
SMILES:	O=C(O)Cc1cc(F)cc(F)c1
Mol. weight [g/mol]:	172.13
CAS:	105184-38-1

Physical Properties

Property code	Value	Unit	Source
gf	-545.73	kJ/mol	Joback Method
hf	-651.89	kJ/mol	Joback Method
hfus	21.59	kJ/mol	Joback Method
hvap	58.79	kJ/mol	Joback Method
log10ws	-2.04		Crippen Method
logp	1.592		Crippen Method
mcvol	110.800	ml/mol	McGowan Method
pc	3810.39	kPa	Joback Method
tb	563.67	K	Joback Method
tc	753.30	K	Joback Method
tf	343.31	K	Joback Method
vc	0.436	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	251.48	J/molxK	563.67	Joback Method
cpg	259.65	J/molxK	595.28	Joback Method
cpg	267.37	J/molxK	626.88	Joback Method
cpg	274.66	J/molxK	658.49	Joback Method
cpg	281.52	J/molxK	690.09	Joback Method
cpg	287.98	J/molxK	721.70	Joback Method
cpg	294.04	J/molxK	753.30	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C105184381&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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