

# Dichloroacetic acid, 3-methylphenyl ester

<b>Inchi:</b>	InChI=1S/C9H8Cl2O2/c1-6-3-2-4-7(5-6)13-9(12)8(10)11/h2-5,8H,1H3
<b>InchiKey:</b>	PIRXAVOHMRJPRX-UHFFFAOYSA-N
<b>Formula:</b>	C9H8Cl2O2
<b>SMILES:</b>	Cc1cccc(OC(=O)C(Cl)Cl)c1
<b>Mol. weight [g/mol]:</b>	219.06

## Physical Properties

Property code	Value	Unit	Source
gf	-132.54	kJ/mol	Joback Method
hf	-285.59	kJ/mol	Joback Method
hfus	20.38	kJ/mol	Joback Method
hvap	56.10	kJ/mol	Joback Method
log10ws	-3.17		Crippen Method
logp	2.704		Crippen Method
mcvol	145.830	ml/mol	McGowan Method
pc	3135.00	kPa	Joback Method
rinqol	1405.00		NIST Webbook
tb	587.69	K	Joback Method
tc	818.84	K	Joback Method
tf	347.13	K	Joback Method
vc	0.547	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	304.59	J/molxK	587.69	Joback Method
cpg	315.62	J/molxK	626.22	Joback Method
cpg	325.90	J/molxK	664.74	Joback Method
cpg	335.46	J/molxK	703.27	Joback Method
cpg	344.32	J/molxK	741.79	Joback Method
cpg	352.48	J/molxK	780.32	Joback Method
cpg	359.97	J/molxK	818.84	Joback Method
dvisc	0.0019276	Paxs	347.13	Joback Method
dvisc	0.0010973	Paxs	387.22	Joback Method

dvisc	0.0006944	Paxs	427.32	Joback Method
dvisc	0.0004752	Paxs	467.41	Joback Method
dvisc	0.0003454	Paxs	507.50	Joback Method
dvisc	0.0002630	Paxs	547.60	Joback Method
dvisc	0.0002078	Paxs	587.69	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U307583&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U307583&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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