

# 2-Pentenoic acid, 4-methyl-, methyl ester

<b>Other names:</b>	Methyl 4-methyl-2-pentenoate
<b>Inchi:</b>	InChI=1S/C7H12O2/c1-6(2)4-5-7(8)9-3/h4-6H,1-3H3/b5-4+
<b>InchiKey:</b>	YVOAQDZUARKSRL-SNAWJCMRSA-N
<b>Formula:</b>	C7H12O2
<b>SMILES:</b>	COC(=O)C=CC(C)C
<b>Mol. weight [g/mol]:</b>	128.17
<b>CAS:</b>	50652-78-3

## Physical Properties

Property code	Value	Unit	Source
gf	-148.08	kJ/mol	Joback Method
hf	-320.67	kJ/mol	Joback Method
hfus	13.35	kJ/mol	Joback Method
hvap	39.90	kJ/mol	Joback Method
log10ws	-1.23		Crippen Method
logp	1.372		Crippen Method
mcvol	112.630	ml/mol	McGowan Method
pc	3142.03	kPa	Joback Method
ripol	1229.00		NIST Webbook
tb	439.57	K	Joback Method
tc	628.84	K	Joback Method
tf	220.73	K	Joback Method
vc	0.425	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	222.78	J/molxK	439.57	Joback Method
cpg	233.88	J/molxK	471.11	Joback Method
cpg	244.50	J/molxK	502.66	Joback Method
cpg	254.66	J/molxK	534.20	Joback Method
cpg	264.37	J/molxK	565.75	Joback Method
cpg	273.65	J/molxK	597.29	Joback Method
cpg	282.49	J/molxK	628.84	Joback Method

dvisc	0.0044540	Paxs	220.73	Joback Method
dvisc	0.0018688	Paxs	257.20	Joback Method
dvisc	0.0009729	Paxs	293.68	Joback Method
dvisc	0.0005850	Paxs	330.15	Joback Method
dvisc	0.0003893	Paxs	366.62	Joback Method
dvisc	0.0002788	Paxs	403.10	Joback Method
dvisc	0.0002111	Paxs	439.57	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C50652783&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C50652783&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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