

# 2,2,2-Trichloroethyl 3-chlorobenzoate

<b>Other names:</b>	Benzoic acid, 3-chloro, 2,2,2-trichloroethyl ester
<b>Inchi:</b>	InChI=1S/C9H6Cl4O2/c10-7-3-1-2-6(4-7)8(14)15-5-9(11,12)13/h1-4H,5H2
<b>InchiKey:</b>	UDOKFBPXDCJNEF-UHFFFAOYSA-N
<b>Formula:</b>	C9H6Cl4O2
<b>SMILES:</b>	O=C(OCC(Cl)(Cl)Cl)c1cccc(Cl)c1
<b>Mol. weight [g/mol]:</b>	287.95

## Physical Properties

Property code	Value	Unit	Source
gf	-151.12	kJ/mol	Joback Method
hf	-320.54	kJ/mol	Joback Method
hfus	24.88	kJ/mol	Joback Method
hvap	63.97	kJ/mol	Joback Method
log10ws	-4.38		Crippen Method
logp	3.867		Crippen Method
mcvol	170.310	ml/mol	McGowan Method
pc	2925.00	kPa	Joback Method
rinpol	1744.00		NIST Webbook
rinpol	1705.00		NIST Webbook
rinpol	1715.00		NIST Webbook
rinpol	1688.00		NIST Webbook
rinpol	1700.00		NIST Webbook
rinpol	1690.00		NIST Webbook
rinpol	1678.00		NIST Webbook
tb	659.76	K	Joback Method
tc	906.78	K	Joback Method
tf	424.39	K	Joback Method
vc	0.640	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	350.48	J/molxK	659.76	Joback Method
cpg	359.71	J/molxK	700.93	Joback Method

cpg	368.06	J/mol×K	742.10	Joback Method
cpg	375.59	J/mol×K	783.27	Joback Method
cpg	382.37	J/mol×K	824.44	Joback Method
cpg	388.45	J/mol×K	865.61	Joback Method
cpg	393.89	J/mol×K	906.78	Joback Method
dvisc	0.0012664	Paxs	424.39	Joback Method
dvisc	0.0007795	Paxs	463.62	Joback Method
dvisc	0.0005175	Paxs	502.85	Joback Method
dvisc	0.0003646	Paxs	542.08	Joback Method
dvisc	0.0002693	Paxs	581.30	Joback Method
dvisc	0.0002066	Paxs	620.53	Joback Method
dvisc	0.0001637	Paxs	659.76	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U373683&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U373683&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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